

LESSON #36: New Smells, New Ideas - Ball-and-Stick Models



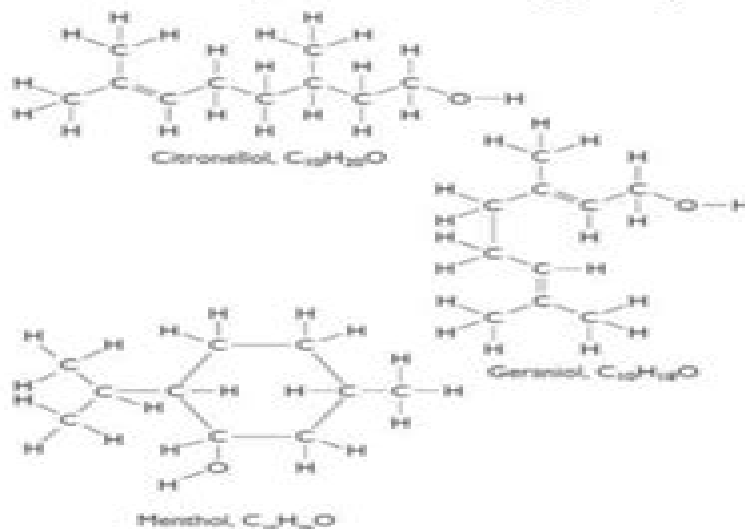
Key Question:

Learning Objectives:

- Interpret three-dimensional ball-and-stick molecular representations
- Translate between molecular models, molecular formulas, and structural formulas
- Describe connections between molecular properties and molecular structure

Key Term: ball-and-stick model

ChemCatalyst: Do you think any of these molecules will smell similar?
What evidence do you have to support your prediction?



Alchemy Unit 2 Smells Lesson 5

Kauṭalya



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Wokesense Conrad Riker, Are you tired of seeing society crumble under the weight of wokesense Are you fed up with the madness of no fault divorce transgenderism and political correctness Are you ready to take a stand and preserve our traditional values This groundbreaking book will help you 1 Understand the sinister origins of wokesense and how it has infected our culture 2 Learn the tactics wokesense uses to divide people and weaken social institutions 3 Discover the true impact of wokesense on families and the abuse of no fault divorce 4 Find out how wokesense pushes absurd gender theories and erases men and women 5 Combat the war on masculinity and rediscover healthy male role models 6 Prepare for the future by embracing the brave world of Red Pill awakening 7 And finally assemble a plan to protect your family and resist wokesense s onslaught If you want to preserve your sanity and protect your family in these crazy times buy this book today

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Amazonia - From Pre-history to Future ,2025-09-03 Six chapters of this book describe recent findings on past and present wildlife and flora in the Amazonia The significance of chemical prospecting in Amazonian biodiversity and biomass dynamics as well as its impact on land uses is presented in two chapters Indigenous and modern perspectives on the use of medicinal plants are also well covered and it is concluded that the protection of forest ecosystems is a major challenge in preserving knowledge about traditional healing methods The potential discovery of new active medicinal molecules explains the broad interest in tropical ethnomedicine Aquaculture development in Brazil has significant potential to reduce rainforest loss by providing a sustainable alternative to traditional livestock farming which is a major driver of deforestation

The Kautīliya Arthaśāstra: An English translation with critical and explanatory notes Kauṭalya,1963 **Warriors** ,2004

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely becuase bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and

explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 M H_2SO_4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0 ml of nitric acid and 4.0 ml of sulfuric acid and will then react with 4.5 ml of bromobenzene this will produce o-bromonitrobenzene

and p bromonitrobenzene

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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