## Analytical Chemistry MCQs: TYBSC SEM III

 The solubility product K<sub>sp</sub> of sparingly soluble Ag<sub>2</sub>CrO<sub>4</sub> is 4\*10<sup>-12</sup>. The molar solubility of the salt is

- (a) 1.0\*10\* moles/lit
- (b) 2.0\*104 moles/lit
- (c) 1.0\*10-5 moles/lit
- (d) 1.2\*10-12 moles/lit

 Solubility product of AgCl is 1.44\*10<sup>-4</sup> moles/lit at 373K. The solubility of the AgCl is boiling of water will be....

- (a) 0.72\*104 moles/lit
- (b) 0.72\*10-2 moles/lit
- (c) 1.20\*10-2 moles/lit
- (d)1.20\*10<sup>-4</sup> moles/lit
- 10. Precipitation takes place when the ionic product...
- (a) Is zero
- (b) Exceeds the solubilty product
- (c) Equal to solubilty product
- (d) Is less than the solubility product
- 11. The electomotive force is expressed In the unit of .....
- (a) Ohm
- (b) Volt
- (c) Ampere
- (d) Colulomb
- 12. If the cell can provide electrical energy to an external system, it is called as
- (a) Galvanic cell
- (b) Electrolytic cell
- (c) Concentration cell
- (d) Chemical cell
- The current per unit area of electrode surface is known as....
- (a) Current efficiency
- (b) Current yield
- (c) Current density
- (d) None of these

# **Analytical Chemistry Questions And Answers**

G. Kateman, L. Buydens

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Quality Control in Analytical Chemistry G. Kateman, L. Buydens, 1993-09-06 Describes the basics of analytical techniques sampling and data handling in order to improve quality control in analytical laboratory management Stresses what quality parameters can be improved and which ones should be rectified first This edition includes numerous modern methods and Basic Analytical Chemistry Mr. Rohit Manglik, 2024-05-18 EduGorilla the latest developments in time proven techniques Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels **Principles of Analytical Chemistry** Miguel Valcarcel, 2012-12-06 Principles of Analytical Chemistry aims to ease the first contact of students pursuing different scientific and technical studies by providing them with a simple general overview of the discipline The objective of this innovative textbook is to teach rather than to inform Using keywords of modern Analytical Chemistry it constructs a teaching message accessible to the beginner with emphasis on relations between these keywords to ensure consistency in the teaching contents of the book Practical exercises 450 questions for students and 27 interactive seminars are used as vehicles for expounding a large number of examples in order to clarify and consolidate the text and facilitate its **Analytical Chemistry** E. Hywel Evans, Mike E. Foulkes, 2019 Analytical Chemistry A Practical Approach comprehension is the only chemical analysis text with an emphasis on active learning giving students step by step guidance on how the key principles of analytical science are applied in a range of practical real world contexts Analytical Chemistry for Technicians, Second Edition John Kenkel, 1994-07-22 The second edition of Analytical Chemistry for Technicians provides the nuts and bolts of analytical chemistry and focuses on the practical aspects for training a technician level laboratory worker This edition presents new and expanded chapters innumerable guestions and problems and modified experiments that present a fresh and challenging approach Some of the topics that have been expanded include chemical equilibrium chromatography Kjeldahl method and molarity and moles where EDTA and water hardness calculations are concerned New discussions of the Aq AqCl and combination pH electrodes have been added while the discussion of ion selective electrodes has been expanded The chapter introducing instrumental analysis and computers now includes discussions of y mx b and the method of least squares The book also includes discussions of FTIR topics of NMR and mass spectrometry which are found in the new infrared spectrometry chapter Handbook of Food Analytical Chemistry, Volume 1 Ronald E. Wrolstad, Terry E. Acree, Eric A. Decker, Michael H. Penner, David S. Reid, Steven J. Schwartz, Charles F. Shoemaker, Denise M. Smith, Peter Sporns, 2005-09-16 Emphasizing effective state of the art methodology and written by recognized experts in the field the Handbook of Food Analytical Chemistry is an indispensable reference for food scientists and technologists to enable successful analysis Provides detailed reports on experimental procedures Includes sections on background theory and

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