# Balancing Chemical Equations – Answer Key

Balance the equations below:

- 1 N<sub>2</sub> + 3 H<sub>2</sub> → 2 NH<sub>3</sub>
- 2) 2 KClO<sub>3</sub> → 2 KCl + 3 O<sub>2</sub>
- 3) 2 NaCl + 1 F<sub>2</sub> → 2 NaF + 1 Cl<sub>2</sub>
- 4) 2 H₂ + 1 O₂ → 2 H₂O
- 5) 1 Pb(OH)<sub>2</sub> + 2 HCl → 2 H<sub>2</sub>O + 1 PbCl<sub>2</sub>
- 6) 2 AlBr<sub>3</sub> + 3 K<sub>2</sub>SO<sub>4</sub> → 6 KBr + 1 Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>
- 7) 1 CH<sub>4</sub> + 2 O<sub>2</sub> → 1 CO<sub>2</sub> + 2 H<sub>2</sub>O
- 8) 1 C<sub>1</sub>H<sub>8</sub> + 5 O<sub>2</sub> → 3 CO<sub>2</sub> + 4 H<sub>2</sub>O
- 9) 2 C<sub>8</sub>H<sub>18</sub> + 25 O<sub>2</sub> → 16 CO<sub>2</sub> + 18 H<sub>2</sub>O
- 10) 1 FeCl<sub>3</sub> + 3 NaOH → 1 Fe(OH)<sub>3</sub> + 3 NaCl
- 11) 4 P + 5 O₂ → 2 P₂O₅
- 12) 2 Na + 2 H<sub>2</sub>O → 2 NaOH + 1 H<sub>2</sub>
- 13) 2 Ag<sub>2</sub>O → 4 Ag + 1 O<sub>2</sub>
- 14) 1 S<sub>8</sub> + 12 O<sub>2</sub> → 8 SO<sub>3</sub>
- 15) 6 CO₂ + 6 H₂O → 1 C<sub>6</sub>H₁₂O<sub>6</sub> + 6 O₂
- 16) 1 K + 1 MgBr → 1 KBr + 1 Mg
- 17) 2 HCl + 1 CaCO<sub>3</sub> → 1 CaCl<sub>2</sub> + 1 H<sub>2</sub>O + 1 CO<sub>2</sub>
- 18) 1 HNO<sub>3</sub> + 1 NaHCO<sub>3</sub> → 1 NaNO<sub>3</sub> + 1 H<sub>2</sub>O + 1 CO<sub>2</sub>
- 19) 2 H<sub>2</sub>O + 1 O<sub>2</sub> → 2 H<sub>2</sub>O<sub>2</sub>
- 20) 2 NaBr + 1 CaF<sub>2</sub> → 2 NaF + 1 CaBr<sub>2</sub>
- 21) 1 H₂SO<sub>4</sub> + 2 NaNO₂ → 2 HNO₂ + 1 Na₂SO<sub>4</sub>

# **Analyzing Chemical Equations Answer Key Only**

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## **Analyzing Chemical Equations Answer Key Only:**

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Qualitative Chemical Analysis Albert Benjamin Prescott, Otis Coe Johnson, 1892 **Encyclopedia of Microfluidics and** Nanofluidics Dongging Li,2008-08-06 Covering all aspects of transport phenomena on the nano and micro scale this encyclopedia features over 750 entries in three alphabetically arranged volumes including the most up to date research insights and applied techniques across all areas Coverage includes electrical double layers optofluidics DNC lab on a chip **GED Test Three** Robert Mitchell,1994 Provides test taking tips practice questions and sample nanosensors and more General Chemistry Laboratory Text with Qualitative Analysis Wilbert Making Skill Standards Work ,1999 tests Hutton, 1968 Nature Sir Norman Lockyer, 1886 **Chemistry** John S. Phillips, Cheryl Wistrom, 2000 Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 Watts' Dictionary of Chemistry Henry Watts, 1890 American Druggists' Circular and Chemical Gazette ,1873 **Numerical Analysis of** Carbon Monoxide Flames by Reduced Kinetic and Asymptotic Methods Maria Lynn Jason Rightley, 1996

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