## 1999 Advanced Placement Chemistry Exam Section I: Multiple Choice

Time — I hour and 30 minutes

#### NO CALCULATORS MAY BE USED WITH SECTION I.

Note: For all questions, assume that the temperature is 298 K, the pressure is 1.00 atmosphere, and solutions are aqueous unless otherwise specified.

Throughout the test the following symbols have the definitions specified unless otherwise noted.

7	- temperature
	- pressure
W	= volume
S	= entropy
	= enthalpy
G	- free energy
R	= molar gas constant
	- number of moles
A16	= molar

488	-	molal
L, mL		liter(s), milliliter(s)
g		gram(s)
nim	100	nanometer(s)
action	100	atmosphere(s)
J, kJ		joule(s), kilojoule(s)
W		volt(s)
med		mole(s)

#### Part A

<u>Directions</u>: Each set of lettered choices below refers to the numbered statements immediately following it. Select the one lettered choice that best fits each statement and then fill in the corresponding oval on the answer sheet. A choice may be used once, more than once, or not at all in each set.

Questions 1-4 refer to the following types of energy.

- (A) Activation energy
- (B) Free energy
- (C) Ionization energy
- (D) Kinetic energy
- (E) Lattice energy
- The energy required to convert a ground-state atom in the gas phase to a gaseous positive ion
- The energy change that occurs in the conversion of an ionic solid to widely separated gaseous ions
- The energy in a chemical or physical change that is available to do useful work
- The energy required to form the transition state in a chemical reaction

Questions 5-8 refer to atoms for which the occupied atomic orbitals are shown below.

- (A) 1s \_ 2s \_
- (B) 1s \_\_ 2s \_\_
- (C) 1s \_ 2s \_ 2p \_ \_ \_
- (D) 1s \_ 2s \_ 2p \_ \_ \_ \_
- (E) [Ar] 4s \_ 3d \_ \_ \_ \_ \_ \_
- 5. Represents an atom that is chemically unreactive
- Represents an atom in an excited state
- 7. Represents an atom that has four valence electrons
- 8. Represents an atom of a transition metal

Questions 9-12 refer to aqueous solutions containing 1:1 mole ratios of the following pairs of substances. Assume all concentrations are 1 M.

# **2008 Ap Chemistry Exam Multiple Choice Answers**

John Michels (Journalist)

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Part III discusses Knowledge Representation in applications such as guestion answering the semantic web automated planning cognitive robotics multi agent systems and knowledge engineering This book is an essential resource for graduate students researchers and practitioners in knowledge representation and AI Make your computer smarter Handle qualitative and uncertain information Improve computational tractability to solve your problems easily Science John Michels Student-staff Directory University of Illinois at Urbana-Champaign, 1995 Modern Techniques for Food Authentication Da-Wen Sun, 2008-06-12 With the increasing awareness of food safety and quality consumers continuously demand the reassurance of origin and content of their foods Furthermore manufacturers must be able to confirm the authenticity of components of their products in order to comply with government legislation Protection of the rights of consumers genuine food processors and prevention of fraudulent or deceptive practices and the adulteration of food is an important and challenge facing the food industry Rapid scientific and technological advances in the determination of food authenticity have taken place in recent years and Modern Techniques for Food Authentication focuses on many of those novel techniques Including coverage of various spectroscopic technologies methods based on isotopic analysis and chromatography DNA enzymatic analysis electrophosresis and thermal methods this book provides a valuable international resource on the latest developments in food authentication A comprehensive overview of authentication techniques and technology Written by an international group of academic and professional peers Provides an excellent complement to more general books on food safety Federal Student Aid George Scott, 2009-12 The Academic Competitiveness AC and National Science and Mathematics Access to Retain Talent SMART Grants were established by the Deficit Reduction Act of 2005 The grants provide merit based financial aid to certain low income college students eligible for Federal Pell Grants and are administered by the Dept of Education DoE In the first year of implementation participation was lower than expected This report determines 1 factors affecting AC and SMART Grant student participation 2 challenges colleges face in administering the grant programs and 3 the extent to which DoE has assisted states and colleges with implementation Includes recommendations Charts and tables **Illustrated Guide to Home Chemistry Experiments** Robert Bruce Thompson, 2008-04-29 Provides information on setting up an in home chemistry lab covers the basics of chemistry and offers a variety of experiments Multiple Solution Methods for Teaching Science in the Classroom Stephen DeMeo, 2008 For the first time in science education the subject of multiple solution methods is explored in book form While a multiple method teaching approach is utilized extensively in math education there are very few journal articles and no texts written on this topic in science Teaching multiple methods to science students in order to solve quantitative word problems is important for two reasons First it challenges the practice by teachers that one specific method should be used when solving problems Secondly it calls into question the belief that multiple methods would confuse students and retard their learning Using a case study approach and informed by research conducted by the author this book claims that providing students with a choice of

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