Analytical Chemistry 3rd Long Exem-

PART 1. MULTIPLE CHOICE. Choose the letter that corresponds to the MEST answer. Write E when the onswer is NOT in the choices, (25 ptc)

1. Transfer of electrons from one expanses to another

a, precialitation run

redox rea

b. bold con-

d. neutrolipotion run

2. What is the male radio of Ce(IV) to NO_V for the oxidation of nitrite to nitrate using Ce++ in the presence of an acid-

a. 1.2 N. 241

6, 141.3 6.00

43

45, 4

3. How many mole(s) of electron lefore) produced in the oxidation of sodium bismuthate (NaBiO-) to BiO+ and Na+ under addic conditions?

a. 1 b. 2

4. Azide (Nr.) was added in the determination of the dissolved axygen in water sample to remove interferences coused by the presence of what lons?

a. dilacida e. ellering to Reportation di. eltroter

- 5. Which of the following statements is/one) true?
 - f. Presoned KWrO., solutions should be stored in amber border
 - II. KNAO, volutions are balled to remove CO; which makes the solution basis
 - III. Mn(II) is the product of the degradation of KMnO.
 - IV. KMnO₄ solutions are filtered to remove management(V) oxide

o. Lonly. c. Food IV b. I and it. 8. L.R. and III. 5. What is the oxidation state of carbon in availa acid?

 $a_1 + 2$ 84. 44 4. 66. 4.77

7. The complex formed between Cuff and 1.10-phenonthroline has a molar absorptivity of 7000 L cm1 mol 1 at 43.5 nm, the wavelength of maximum absorption. What is the absorbance of a 6.17 x 10-1 M solution of the complex when measured in a 1,00-on cell of 435 and

a. 0.0431 6.0.0432 4.0.432 d. 0.431

8. In the presence of lodine, thiosulfate ion is axidiped quantitatively to tetrohionate Loss, Webcot Is that obsessional formulas and testmostatements loss?

a. 5/Ov 4.5000 by S. Ohn 4.540-

9. In Winkler ritration, the floc is produced from

as exidentee of It to by

e, reclusting of I to Iv-

b. exidetion of Mn(III) to Mn(III)

d. reduction of Mn(II) to Mn(III)

10. A component of a UV-Vis spectrophotometer that restricts the radiation being measured to a narrow band that is obsorbed or enlitted by the analyte.

or Boths source:

c. rediction filter

In detector

d. wavelength selector

 Component of a UV-Vis instrument that operates by blacking or obsorbing oil but a resiriesed band of radiation.

a. Saht source:

c. radiation filter.

h. detector

d, wavelength selector

 Oxidation of VO!" using permanganate or cerium as standard solution produces. VIONLY, what is the axidation state of vanadium in VIOHL+7.

Analytical Chemistry Exam Questions

Kim Seng Chan, Jeanne Tan

Analytical Chemistry Exam Questions:

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for determining the excellence of quantitative assays available for analysts to utilize for method validation This book discusses as well the primary challenge of neuropharmacologists to relate physiological functions to the many ligand binding sites identified in brain tissue The final chapter deals with the fundamentals and applications of biosensors This book is a valuable resource for analytical chemists chemical engineers clinical chemists neuropharmacologists and scientists Level Chemistry Questions and Answers PDF Arshad Igbal, The A Level Chemistry Quiz Questions and Answers PDF IGCSE GCE Chemistry Competitive Exam Questions Chapter 1 28 Practice Tests Class 11 12 Chemistry Textbook Questions for Beginners includes revision guide for problem solving with hundreds of solved questions A Level Chemistry Questions and Answers PDF book covers basic concepts analytical and practical assessment tests A Level Chemistry Quiz PDF book helps to practice test questions from exam prep notes The A Level Chemistry Quiz Questions and Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved tests A Level Chemistry Questions and Answers PDF Free download chapter 1 a book covers solved common questions and answers on chapters Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry ionic equilibria lattice energy moles and equations nitrogen and sulfur organic and nitrogen compounds periodicity polymerization rates of reaction reaction kinetics redox reactions and electrolysis states of matter transition elements tests for college and university revision guide Chemistry Interview Questions and Answers PDF Download free eBook's sample covers beginner's solved questions textbook's study notes to practice online tests The IGCSE GCE Chemistry Interview Questions Chapter 1 28 PDF book includes high school question papers to review practice tests for exams A Level Chemistry Practice Tests a textbook s revision guide with chapters tests for IGCSE NEET MCAT GRE GMAT SAT ACT competitive exam A Level Chemistry Questions Bank Chapter 1 28 PDF book covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Alcohols and Esters Questions Chapter 2 Atomic Structure and Theory Questions Chapter 3 Benzene Chemical Compound Questions Chapter 4 Carbonyl Compounds Questions Chapter 5 Carboxylic Acids and Acyl Compounds Questions Chapter 6 Chemical Bonding Questions Chapter 7 Chemistry of Life Questions Chapter 8 Electrode Potential Questions Chapter 9 Electrons in Atoms Questions Chapter 10 Enthalpy Change Questions Chapter 11 Equilibrium Questions Chapter 12 Group IV Questions Chapter 13 Groups II and VII Questions Chapter 14 Halogenoalkanes Questions Chapter 15 Hydrocarbons Questions Chapter 16 Introduction to Organic Chemistry Questions Chapter 17 Ionic Equilibria Questions Chapter 18 Lattice Energy Questions Chapter 19 Moles and Equations Questions Chapter 20 Nitrogen and Sulfur Questions Chapter 21 Organic and Nitrogen Compounds Questions Chapter 22 Periodicity Questions Chapter 23 Polymerization Questions Chapter 24 Rates of Reaction Questions Chapter 25 Reaction Kinetics Questions Chapter 26 Redox Reactions and Electrolysis Questions Chapter 27 States

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