

Unit 7 Atmospheric Pollution APES Exam Review

Outdoor Air Pollutants

- List ALL the air pollutants emitted from burning coal.
 - CO_2 , CO , SO_2 , NO_x , particulates, Heavy metals (ie- Mercury and Lead)
- Nitrogen oxides are primary air pollutants. However, they can mix with other compounds in the air to make photochemical smog and acid rain. Smog and acid rain are examples of what type of pollutants?
 - Secondary air pollutants
- The burning of diesel fuel releases what type of outdoor air pollutant?
 - Lots of particulate matter ("soot"), hydrocarbons, CO_2 , CO , NO_x



Photochemical Smog

- In the box below, write a series of chemical reactions that leads to the formation of tropospheric ozone in photochemical smog.

Atmospheric oxidant production:

- $\text{NO} + \text{VOC} \longrightarrow \text{NO}_2$ (nitrogen dioxide)
- $\text{NO}_2 + \text{UV} \longrightarrow \text{NO} + \text{O}$ (nitric oxide + atomic oxygen)
- $\text{O} + \text{O}_2 \longrightarrow \text{O}_3$ (ozone)
- $\text{NO}_2 + \text{VOC} \longrightarrow \text{PAN, etc.}$ (peroxyacetyl nitrate)

Net results:



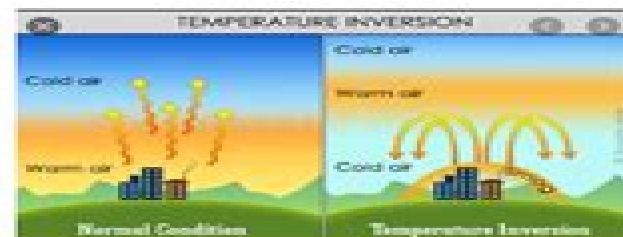
- What is a VOC? What are their main sources?
 - Volatile Organic Compounds are released into the air from solids and liquids that evaporate quickly/easily.
 - Main sources of VOC pollution include: paint, gasoline, cleaning products
- In what seasons do we see increases in smog concentration? Why?
 - Summer because photochemical smog creation reactions require sunlight
- Why are there high levels of smog in urban areas?
 - There are more VOC's from cars and industry. (VOC's are a smog formation reactant)

Thermal Inversion

- In the two boxes below, draw what normal conditions in the atmosphere look like. Then draw what thermal inversion conditions look like in the atmosphere.

- Why do thermal inversion layers trap pollutants?

The warm air layer traps smog near the surface instead of allowing it to dissipate upwards or away from human living areas.



Particulate Matter

- What are natural sources of particulate matter in the air? Rock dust, dust storms, volcanic eruptions, naturally occurring forest fires (ex- due to lightning)
- What's the difference between PM 2.5 and PM 10? PM 2.5 particles are much smaller than PM 10. Smaller particles are very dangerous because they have a better chance of sneaking past our filtration system (nose, mucous, etc) into the lungs.

Carbon Dioxide

- What are some natural sources of carbon dioxide in the atmosphere? Ocean-atmosphere exchange, Cell respiration of ALL organisms (plants AND animals do cell resp)

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