

Experiment 17 Report Sheet

Antacid Analysis

Date _____ Lab Sec. _____ Name _____ Desk No. _____

A. Dissolving the Antacid

	Trial 1	Trial 2	Trial 1	Trial 2
1. Mass of flask (g)	128.457	97.342	115.395	111.220
2. Mass of flask + antacid sample (g)	128.661	97.542	115.606	111.440
3. Mass (or tared mass) of antacid sample (g)	.204	.200	0.211	0.217
4. Total volume of HCl added (mL)	25	25	25	25
5. Molar concentration of HCl (mol/L)	.09687		0.9687	

B. Analyzing the Antacid Sample

1. Molar concentration of NaOH (mol/L)	.09361		0.9361	
2. Buret reading, initial (mL)	1.5	16.7	9.7	11.6
3. Buret reading, final (mL)	16.7	28.4	18.2	21.7
4. Volume of NaOH (mL)	15.2	12.3	8.5	10.1

C. Calculations

1. Moles of HCl added, total (mol)				
2. Moles of NaOH added (mol)				
3. Moles of base in antacid sample (mol)				
4. $\frac{\text{mol base in antacid}}{\text{mass of antacid sample}}$ (mol/g)		*		*
5. Average $\frac{\text{mol base in antacid}}{\text{mass of antacid sample}}$ (mol/g)				

*Show calculation(s) for Trial(s) 1 on the next page.

Analysis Of Antacids Lab Report Answers

Allison Hale, Mary Jo Hovey



Analysis Of Antacids Lab Report Answers:

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Table of Contents Analysis Of Antacids Lab Report Answers

1. Understanding the eBook Analysis Of Antacids Lab Report Answers
 - The Rise of Digital Reading Analysis Of Antacids Lab Report Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Analysis Of Antacids Lab Report Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Analysis Of Antacids Lab Report Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Analysis Of Antacids Lab Report Answers
 - Personalized Recommendations
 - Analysis Of Antacids Lab Report Answers User Reviews and Ratings
 - Analysis Of Antacids Lab Report Answers and Bestseller Lists
5. Accessing Analysis Of Antacids Lab Report Answers Free and Paid eBooks
 - Analysis Of Antacids Lab Report Answers Public Domain eBooks
 - Analysis Of Antacids Lab Report Answers eBook Subscription Services
 - Analysis Of Antacids Lab Report Answers Budget-Friendly Options

6. Navigating Analysis Of Antacids Lab Report Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Analysis Of Antacids Lab Report Answers Compatibility with Devices
 - Analysis Of Antacids Lab Report Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Analysis Of Antacids Lab Report Answers
 - Highlighting and Note-Taking Analysis Of Antacids Lab Report Answers
 - Interactive Elements Analysis Of Antacids Lab Report Answers
8. Staying Engaged with Analysis Of Antacids Lab Report Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Analysis Of Antacids Lab Report Answers
9. Balancing eBooks and Physical Books Analysis Of Antacids Lab Report Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Analysis Of Antacids Lab Report Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Analysis Of Antacids Lab Report Answers
 - Setting Reading Goals Analysis Of Antacids Lab Report Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Analysis Of Antacids Lab Report Answers
 - Fact-Checking eBook Content of Analysis Of Antacids Lab Report Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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