Analytical Chemistry MCQs: TYBSC SEM III

8. The solubility product K_{sp} of sparingly soluble Ag_2CrO_4 is $4*10^{-12}$. The molar solubility of the salt is

(a) $1.0*10^{-4}$ moles/lit

(b) $2.0*10^{-4}$ moles/lit

(c) $1.0*10^{-5}$ moles/lit

(d) $1.2*10^{-12}$ moles/lit

9. Solubility product of AgCl is $1.44*10^{-4}$ moles/lit at 373K. The solubility of the AgCl is boiling of water will be....

(a) $0.72*10^{-4}$ moles/lit

(b) $0.72*10^{-2}$ moles/lit

(c) $1.20*10^{-2}$ moles/lit

(d)$1.20*10^{-4}$ moles/lit

10. Precipitation takes place when the ionic product...

(a) Is zero

(b) Exceeds the solubilty product

(c) Equal to solubilty product

(d) Is less than the solubility product

11. The electomotive force is expressed In the unit of .....

(a) Ohm

(b) Volt

(c) Ampere

(d) Colulomb

12. If the cell can provide electrical energy to an external system, it is called as

(a) Galvanic cell

(b) Electrolytic cell

(c) Concentration cell

(d) Chemical cell

13. The current per unit area of electrode surface is known as....

(a) Current efficiency

(b) Current yield

(c) Current density

(d) None of these

Analytical Chemistry Questions And Answers

Annelies Wilder-Smith

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