

2010 GMC Yukon Denali/Yukon XL Denali Owner Manual

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2008 Gmc Yukon Denali Owners Manual

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temperature change plugging in the values given in the problem you get $q_{\text{water}} = 418 \text{ J/g} \cdot \text{C} \times 110 \text{ g} \times 26.6 \text{ C} = 11900 \text{ J}$
 $550 \text{ J} = \Delta h \cdot q_{\text{water}} = 550 \text{ J}$

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 for water $4184 \text{ J/g} \cdot \text{C}$ 1 what is the specific heat of aluminum if the temperature of a 28.4 g sample of aluminum is increased
 by 8.1 C when 207 J of heat is added 2

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 $147314 \Delta H_{\text{rxn}} = n \cdot q_{\text{rxn}} = n \cdot q_{\text{calorimeter}} = m \cdot c \cdot \Delta T$

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 of 1.00 M NaOH at 25.0 C in a foam cup calorimeter a reaction occurs and the temperature rises to 32.0 C calculate the
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