

13. (4 pts) What is an ionic compound?
- A compound that will dissolve in water
 - Two nonmetal atoms bonded together
 - Two metal atoms bonded together
 - A metal and a nonmetal atom bonded together
 - None of the above
14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
- Ionic
 - Covalent
 - Both a and b
 - Neither a nor b
15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
- Ionic
 - Covalent
 - Both a and b
 - Neither a nor b
16. (4 pts) How many total atoms are in this molecule: $C_6H_{12}OH$?
- 3
 - 4
 - 19
 - 20
 - None of the above
17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is $C_5H_{10}Cl_x$, what must x be?
- 1
 - 2
 - 3
 - 4
 - None of the above
18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water? C_3H_6ClF $C_6H_5CH_3$ CS_2 $BaSO_4$
- C_3H_6ClF
 - $C_6H_5CH_3$
 - CS_2
 - $BaSO_4$
 - CS_2 and $BaSO_4$

Answers For Chemistry Module 13

Padhraic Smyth



Answers For Chemistry Module 13:

Nelson Modular Science Paul Collinson, Paul Collison, 2005-03 The Nelson Modular Science series is made up of three books divided into single double and triple award modules presented in an accessible format Book 1 covers the six single award and one coursework modules Book 2 contains six double award modules and Book 3 covers the six triple award modules Each module is covered in self contained units This teacher s file includes practical support sheets and addresses Sc1 investigations Works sheets are provided to integrate the use of ICT throughout science Additional GCSE style questions and modular tests should enhance learning and recall of information *NEET UG Chemistry Study Notes with Theory + Practice MCQs for Complete Preparation | Based on New Syllabus as per NMC* EduGorilla Prep Experts, *AS Chemistry for AQA* John Atkinson, Carol Hibbert, 2000 This chemistry text is written to match exactly the specification for teaching Advanced Chemistry from September 2000 There are two strands AS and A2 with student books The accompanying resource packs are also available on CD ROM Chemistry John Atkinson, Carol Hibbert, 2001 This chemistry text is written to match exactly the specification for teaching Advanced Chemistry from September 2000 There are two strands AS and A2 with student books The accompanying resource packs are also available on CD ROM *Cambridge IGCSE® Chemistry Revision Guide* Roger Norris, 2015-10-08 The Cambridge IGCSE Chemistry Revision Guide supports students through their course containing specifically designed features to help students apply their knowledge as they prepare for assessment This Revision Guide offers support for students as they prepare for their Cambridge IGCSE Chemistry 0620 exams Containing up to date material that matches the syllabus for examination from 2016 and packed full of guidance such as Worked Examples Tips and Progress Check questions throughout to help students to hone their revision and exam technique and avoid common mistakes These features have been specifically designed to help students apply their knowledge in exams Written in a clear and straightforward tone this Revision Guide is perfect for international learners **JEE Main Chemistry - Unit wise Practice Test Papers** Career Point Kota, 2020-07-19 Competitive examination preparation takes enormous efforts time on the part of a student to learn practice and master each unit of the syllabus To check proficiency level in each unit student must take self assessment to identify his her weak areas to work upon that eventually builds confidence to win Also performance of a student in exam improves significantly if student is familiar with the exact nature type and difficulty level of the questions being asked in the Exam With this objective in mind we are presenting before you this book containing unit tests Some features of the books are The complete syllabus is divided into logical units and there is a self assessment tests for each unit Tests are prepared by subject experts who have decade of experience to prepare students for competitive exams Tests are as per the latest pattern of the examination Detailed explanatory solution of each test paper is also given Student is advised to attempt these Tests once they complete the preparation revision of unit They should attempt these Test in exam like environment in a specified time Student is advised to properly analyze the solutions and think of alternative

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EPA Publications Bibliography Quarterly Abstract Bulletin United States. Environmental Protection Agency, 1999-04

Whitaker's Books in Print, 1998

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