

e. Pulmonary embolism

24. Drug metabolism

- a) Polar molecules cross mucous membrane easily
- b) Drug metabolism tends to make molecules more polar
- c) A base is less ionized in an acidic environment

This is a toxicology question in disguise! Has to do with enhanced elimination in toxicology.

Metabolism tends to make lipophilic drugs more polar (more excretable by the kidney) as ionized drug aren't reabsorbed.

Bases (ie most drugs), more ionized in acid environment so they are 'trapped' in acid urine and not reabsorbed. And acid drugs are more ionized in alkaline environment

Polar molecules tend not to cross mucous membranes

3. Formation of acetone occurs in poisoning with:

- a) ethanol
- b) methanol
- c) Ethyleneglycol ?
- d) isopropanol
- e) acetaminophen x

12. The following blood gases pH 7.17, PCO₂ 59, Bic 21, PO₂ 130 are most

likely to be consistent with:

- a) diabetic ketoacidosis
- b) diuretic overdose
- c) premature twin baby
- d) camphor ingestion with seizures
- e) oliguria and renal failure

15. One of the differences between morphine and pethidine relates to:

- a) analgesic efficacy
- b) route of metabolism
- c) ability to suppress cough
- d) histamine release effects
- e) abuse potential ?

68. All of the following substances bind well to activated charcoal EXCEPT:

- a) thioridazine
- b) atenolol
- c) cyanide
- d) benzotropine
- e) tetrahydrocannabinol (THC)

69. In paracetamol poisoning:

- a) there are theoretical advantages to giving the antidote at 2 hours post ingestion
- b) toxicity is less likely in children than adults
- c) the initial dose of N-acetylcysteine is given over 5 minutes
- d) N-acetylcysteine is not indicated more than 20 hours post ingestion
- e) activated charcoal is not used more than 1 hour after ingestion

Answered Mcq In Toxicology

R. N. Karmakar



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Table of Contents Answered Mcq In Toxicology

1. Understanding the eBook Answered Mcq In Toxicology
 - The Rise of Digital Reading Answered Mcq In Toxicology
 - Advantages of eBooks Over Traditional Books
2. Identifying Answered Mcq In Toxicology
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answered Mcq In Toxicology
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answered Mcq In Toxicology
 - Personalized Recommendations
 - Answered Mcq In Toxicology User Reviews and Ratings
 - Answered Mcq In Toxicology and Bestseller Lists
5. Accessing Answered Mcq In Toxicology Free and Paid eBooks
 - Answered Mcq In Toxicology Public Domain eBooks
 - Answered Mcq In Toxicology eBook Subscription Services
 - Answered Mcq In Toxicology Budget-Friendly Options

6. Navigating Answered Mcq In Toxicology eBook Formats
 - ePub, PDF, MOBI, and More
 - Answered Mcq In Toxicology Compatibility with Devices
 - Answered Mcq In Toxicology Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answered Mcq In Toxicology
 - Highlighting and Note-Taking Answered Mcq In Toxicology
 - Interactive Elements Answered Mcq In Toxicology
8. Staying Engaged with Answered Mcq In Toxicology
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answered Mcq In Toxicology
9. Balancing eBooks and Physical Books Answered Mcq In Toxicology
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answered Mcq In Toxicology
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answered Mcq In Toxicology
 - Setting Reading Goals Answered Mcq In Toxicology
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answered Mcq In Toxicology
 - Fact-Checking eBook Content of Answered Mcq In Toxicology
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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