

Grade 11 Chemistry

INTERMOLECULAR FORCES & Physical Properties

EXAM QUESTIONS PART 3



NAME OF SUBSTANCE	FORMULA	BOILING POINT (°C)
Water	H ₂ O	100
Ethanol	CH ₃ CH ₂ OH	78
Bromine	Br ₂	58,8
Iodine	I ₂	184,3
Ammonia	NH ₃	-33,3
Phosphine	PH ₃	-87,7

Explain why ethanol is soluble in water. Refer to the relative strength of the intermolecular forces in ethanol and water.

Explain why the boiling point of iodine is higher than that of bromine. Refer to the intermolecular forces present in EACH substance in the explanation.

Answers For Grade 11 Intermolecular Formal Task

L Darling-Hammond



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Table of Contents Answeres For Grade 11 Intermolecular Formal Task

1. Understanding the eBook Answeres For Grade 11 Intermolecular Formal Task
 - The Rise of Digital Reading Answeres For Grade 11 Intermolecular Formal Task
 - Advantages of eBooks Over Traditional Books
2. Identifying Answeres For Grade 11 Intermolecular Formal Task
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answeres For Grade 11 Intermolecular Formal Task
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answeres For Grade 11 Intermolecular Formal Task
 - Personalized Recommendations

- Answeres For Grade 11 Intermolecular Formal Task User Reviews and Ratings
- Answeres For Grade 11 Intermolecular Formal Task and Bestseller Lists
- 5. Accessing Answeres For Grade 11 Intermolecular Formal Task Free and Paid eBooks
 - Answeres For Grade 11 Intermolecular Formal Task Public Domain eBooks
 - Answeres For Grade 11 Intermolecular Formal Task eBook Subscription Services
 - Answeres For Grade 11 Intermolecular Formal Task Budget-Friendly Options
- 6. Navigating Answeres For Grade 11 Intermolecular Formal Task eBook Formats
 - ePub, PDF, MOBI, and More
 - Answeres For Grade 11 Intermolecular Formal Task Compatibility with Devices
 - Answeres For Grade 11 Intermolecular Formal Task Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answeres For Grade 11 Intermolecular Formal Task
 - Highlighting and Note-Taking Answeres For Grade 11 Intermolecular Formal Task
 - Interactive Elements Answeres For Grade 11 Intermolecular Formal Task
- 8. Staying Engaged with Answeres For Grade 11 Intermolecular Formal Task
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answeres For Grade 11 Intermolecular Formal Task
- 9. Balancing eBooks and Physical Books Answeres For Grade 11 Intermolecular Formal Task
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answeres For Grade 11 Intermolecular Formal Task
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answeres For Grade 11 Intermolecular Formal Task
 - Setting Reading Goals Answeres For Grade 11 Intermolecular Formal Task
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answeres For Grade 11 Intermolecular Formal Task
 - Fact-Checking eBook Content of Answeres For Grade 11 Intermolecular Formal Task

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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