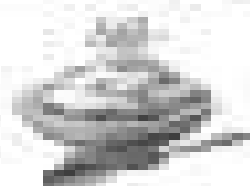


Atomic Probing Structure of the Atom

Name _____
Date _____



Purpose:

To determine which elements are present in the gas by observing the spectrum of light it emits.

Part I: Making Observations in the Atomic Probing

1. Turn on the power supply to the atomic probe. The gas in the tube will begin to glow. Observe the color of the light emitted. Record your observations in the table below. (The color of the light emitted is a characteristic of the element present in the gas.)

Observation: The gas in the tube glows with a bright yellow-orange light. (This is the characteristic color of sodium.)

2. Turn on the power supply to the atomic probe. Observe the color of the light emitted. Record your observations in the table below.

Observation: The gas in the tube glows with a bright blue-green light. (This is the characteristic color of mercury.)

3. Turn on the power supply to the atomic probe. Observe the color of the light emitted. Record your observations in the table below.

Observation: The gas in the tube glows with a bright red-orange light. (This is the characteristic color of hydrogen.)

4. Turn on the power supply to the atomic probe. Observe the color of the light emitted. Record your observations in the table below.

Answers Lesson 11 Atomic Pudding

RS Peters



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