

Chemical Formula Writing Worksheet Solutions

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl₂", as shown.

	<i>zinc</i>	<i>iron (II)</i>	<i>iron (III)</i>	<i>gallium</i>	<i>silver</i>	<i>lead (IV)</i>
<i>chloride</i>	ZnCl₂	FeCl₂	FeCl₃	GaCl₃	AgCl	PbCl₄
<i>acetate</i>	Zn(C₂H₃O₂)₂	Fe(C₂H₃O₂)₂	Fe(C₂H₃O₂)₃	Ga(C₂H₃O₂)₃	Ag C₂H₃O₂	Pb(C₂H₃O₂)₄
<i>nitrate</i>	Zn(NO₃)₂	Fe(NO₃)₂	Fe(NO₃)₃	Ga(NO₃)₃	AgNO₃	Pb(NO₃)₄
<i>oxide</i>	ZnO	FeO	Fe₂O₃	Ga₂O₃	Ag₂O	PbO₂
<i>nitride</i>	Zn₃N₂	Fe₃N₂	FeN	GaN	Ag₃N	Pb₃N₄
<i>sulfate</i>	ZnSO₄	FeSO₄	Fe₂(SO₄)₃	Ga₂(SO₄)₃	Ag₂SO₄	Pb(SO₄)₂

Write the formulas for the following compounds:

- 1) copper (II) chloride **CuCl₂**
- 2) lithium acetate **LiC₂H₃O₂**
- 3) vanadium (III) selenide **V₂Se₃**
- 4) manganese (IV) nitride **Mn₃N₄**
- 5) beryllium oxide **BeO**
- 6) sodium sulfate **Na₂SO₄**
- 7) aluminum arsenide **AlAs**
- 8) potassium permanganate **KMnO₄**
- 9) chromium (VI) cyanide **Cr(CN)₆**
- 10) tin (II) sulfite **SnSO₃**
- 11) vanadium (V) fluoride **VF₅**
- 12) ammonium nitrate **NH₄NO₃**

Answers For Ionic Formula Writing

Léo Pomar

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Answers For Ionic Formula Writing

1. Understanding the eBook Answers For Ionic Formula Writing
 - The Rise of Digital Reading Answers For Ionic Formula Writing
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Ionic Formula Writing
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Ionic Formula Writing
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Ionic Formula Writing
 - Personalized Recommendations
 - Answers For Ionic Formula Writing User Reviews and Ratings
 - Answers For Ionic Formula Writing and Bestseller Lists
5. Accessing Answers For Ionic Formula Writing Free and Paid eBooks
 - Answers For Ionic Formula Writing Public Domain eBooks
 - Answers For Ionic Formula Writing eBook Subscription Services
 - Answers For Ionic Formula Writing Budget-Friendly Options

6. Navigating Answers For Ionic Formula Writing eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Ionic Formula Writing Compatibility with Devices
 - Answers For Ionic Formula Writing Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Ionic Formula Writing
 - Highlighting and Note-Taking Answers For Ionic Formula Writing
 - Interactive Elements Answers For Ionic Formula Writing
8. Staying Engaged with Answers For Ionic Formula Writing
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Ionic Formula Writing
9. Balancing eBooks and Physical Books Answers For Ionic Formula Writing
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Ionic Formula Writing
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers For Ionic Formula Writing
 - Setting Reading Goals Answers For Ionic Formula Writing
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers For Ionic Formula Writing
 - Fact-Checking eBook Content of Answers For Ionic Formula Writing
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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