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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$ reduction $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$ 24 3 $\text{Ag} + \text{Ni} \rightarrow \text{Ag} + \text{Ni}$ 3 $\text{Ag} + \text{Ni} \rightarrow \text{Ag} + \text{Ni}$ 3
oxidation reduction 25 $\text{Cu} + \text{Pb} \rightarrow \text{Cu} + \text{Pb}$ 26 O_2

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 $\text{I}_2 + 5\text{CO} \rightarrow 2\text{IO}_3 + 5\text{CO}_2$

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