

THEORY

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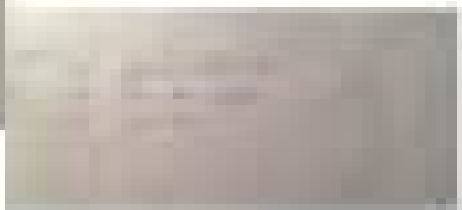
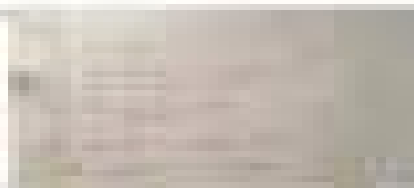
1. Introduction

The purpose of this experiment is to determine the rate of reaction between hydrogen peroxide and potassium iodide. The reaction is catalyzed by a small amount of potassium iodide. The rate of reaction is measured by the volume of oxygen gas produced over a period of time. The reaction is exothermic, and the temperature of the reaction mixture is also measured.

The reaction is as follows:

$$2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$$

The rate of reaction is determined by measuring the volume of oxygen gas produced over a period of time. The volume of oxygen gas is measured using a gas syringe.



Algebra 2 Unit 07 Lesson 01 Just Like Fractions

A. A. Frempong



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