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Instructor's Solutions Manual
to accompany

ANALYSIS

with an Introduction
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5th Edition

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Analysis Steven Lay 4th Edition Solutions Manual

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 $2 \text{ NaCl} + \text{NaCl} \rightarrow 3 \text{ Al} + 2 \text{ Al}_2\text{O}_3 + 4 \text{ Al}_2\text{O}_3 + 2 \text{ Al}_2\text{O}_3 + 4 \text{ N}_2 + 2 \text{ H}_2 + 3 \text{ N}_2$

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 at the molecular level the reaction would look something like this notice that there are two oxygen atoms on the

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 $\text{C}_3\text{H}_8 + 2 \text{ H}_2\text{O} \rightarrow 3 \text{ CO}_2$