# AQA A LEVEL PHYSICS PAPER 1

- What is meant by work function (2): Minimum energy required by an electron to escape from (1) a metal surface (1)
- What is meant by ionisation energy? (2): Minimum energy required to remove an electron from an atom (1) from the ground state (1)
- 3. Why do emitted electrons have a kinetic energy that varies up to a maximum value? (4): Energy of a photon is fixed/constant (1)

  Energy is required for electrons to overcome work function Ø (1)

  Electrons deeper into/below the surface of the metal require more energy to over- come work function and therefore have less KE(1)

  Maximum KE = Energy of photon work function Ø (1)
- 4. Explain how (bright ring) diagram is evidence of the wave-particle duality of electrons?: Diffraction pattern maxima/minima (bright and dark areas) (1) Diffraction is a property of waves- proves wave-behaviour of electrons (1)
- 5. What is meant by the ground state?: The lowest energy state of an atom 6. Explain the difference between excitation and ionisation (2): An electron receives a discrete amount of energy for BOTH (1) Excitation promotes an electron to a higher energy level lonisation is when an electron receives enough energy to leave the atom 7. Explain why only photons of certain frequencies cause excitation (4):
- dis- crete energy levels (1)
  - Electrons need to absorb a DISCRETE amount of energy to move to a higher energy level (1)

# **Aqa As Physics Empa 2written Paper**

Jeremy Pollard, Carol Davenport, Nicky Thomas, Nick England

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# Table of Contents Aqa As Physics Empa 2written Paper

- 1. Understanding the eBook Aqa As Physics Empa 2written Paper
  - The Rise of Digital Reading Aqa As Physics Empa 2written Paper
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Aqa As Physics Empa 2written Paper
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Aga As Physics Empa 2written Paper
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Aga As Physics Empa 2written Paper
  - Personalized Recommendations
  - Aga As Physics Empa 2written Paper User Reviews and Ratings
  - Aga As Physics Empa 2written Paper and Bestseller Lists

- 5. Accessing Aga As Physics Empa 2written Paper Free and Paid eBooks
  - Aga As Physics Empa 2written Paper Public Domain eBooks
  - Aga As Physics Empa 2written Paper eBook Subscription Services
  - Aga As Physics Empa 2written Paper Budget-Friendly Options
- 6. Navigating Aga As Physics Empa 2written Paper eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Aga As Physics Empa 2written Paper Compatibility with Devices
  - Aqa As Physics Empa 2written Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Aqa As Physics Empa 2written Paper
  - Highlighting and Note-Taking Aqa As Physics Empa 2written Paper
  - o Interactive Elements Aqa As Physics Empa 2written Paper
- 8. Staying Engaged with Aqa As Physics Empa 2written Paper
  - o Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Aqa As Physics Empa 2written Paper
- 9. Balancing eBooks and Physical Books Aqa As Physics Empa 2written Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Aga As Physics Empa 2written Paper
- 10. Overcoming Reading Challenges
  - o Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Aqa As Physics Empa 2written Paper
  - Setting Reading Goals Aga As Physics Empa 2written Paper
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Aga As Physics Empa 2written Paper
  - Fact-Checking eBook Content of Aqa As Physics Empa 2written Paper
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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