

Diffusion

Particles move about randomly, and after a bit they end up evenly spaced. It's not rocket science, is it...

Don't Be Put Off by the Fancy Word

- 1) "Diffusion" is simple. It's just the gradual movement of particles from places where there are lots of them to places where there are fewer of them — it's just the natural tendency for stuff to spread out.
- 2) Unfortunately you also have to learn the fancy way of saying the same thing, which is this:

DIFFUSION is the SPREADING OUT of particles from an area of HIGHER CONCENTRATION to an area of LOWER CONCENTRATION.

- 4) Diffusion happens in both solutions and gases — that's because the particles in these substances are free to move about randomly.
- 5) The simplest type is when different gases diffuse through each other. This is what's happening when the smell of perfume diffuses through the air in a room:



- 6) The bigger the concentration gradient (the difference in concentration), the faster the diffusion rate.
- 7) A higher temperature will also give a faster diffusion rate because the particles have more energy, so move around faster.

Cell Membranes Are Kind of Clever...

- 1) They're clever because they hold the cell together BUT they let stuff in and out as well.
- 2) Dissolved substances can move in and out of cells by diffusion.
- 3) Only very small molecules can diffuse through cell membranes though — things like oxygen (needed for respiration — see page 62), glucose, amino acids and water.
- 4) Big molecules like starch and proteins can't fit through the membrane:



- 5) Just like with diffusion in air, particles flow through the cell membrane from where there's a higher concentration (a lot of them) to where there's a lower concentration (not such a lot of them).
- 6) They're only moving about randomly of course, so they go both ways — but if there are a lot more particles on one side of the membrane, there's a net (overall) movement from that side.
- 7) The larger the surface area of the membrane, the faster the diffusion rate, because more particles can pass through at once — see page 23.

Revision by diffusion — you wish...

Wouldn't it be great if all the ideas in this book would just gradually drift across into your mind...

- Q1 A student adds a drop of ink to a glass of cold water.
- a) What will the student observe to happen to the drop of ink. Explain your answer. [2 marks]
 - b) How might the observation differ if the ink was added to a glass of warm water? [1 mark]

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R Barnett



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