

THEORY

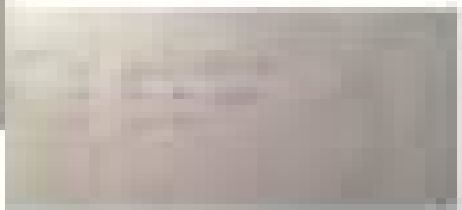
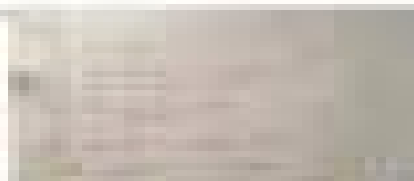
EXPERIMENT

1. Introduction

The purpose of this experiment is to determine the rate of reaction between hydrogen peroxide and potassium iodide. The reaction is catalyzed by the presence of a small amount of potassium iodide. The rate of reaction is measured by the volume of oxygen gas evolved over a period of time.

2. Apparatus and Materials

- Hydrogen peroxide solution (3%)
- Potassium iodide solution (0.1M)
- Concentrated sulfuric acid
- Distilled water
- Conical flask
- Delivery tube
- Gas syringe
- Stopwatch



Algebra 2 Answers Unit 07 Lesson 01

Scott C. Dulebohn

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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