

- 1 Amended from AQA Chemistry Unit 1 Foundation Chemistry CHEM1 June 2010
(Question 1)

Fluorine forms many compounds that contain covalent bonds.

- a i State the meaning of the term *covalent bond*.

.....
.....

(1 mark)

- ii Write an equation to show the formation of one molecule of ClF_3 from chlorine and fluorine molecules.

.....
.....

(1 mark)

- b Draw the shape of a dichlorodifluoromethane molecule (CCl_2F_2) and the shape of a chlorine trifluoride molecule (ClF_3). Include any lone pairs of electrons that influence the shape.

Shape of CCl_2F_2

Shape of ClF_3

(2 marks)

- c Suggest the strongest type of intermolecular force between CCl_2F_2 molecules.

.....
.....

(1 mark)

- d BF_3 is a covalent molecule that reacts with an F^- ion to form a BF_4^- ion.

- i Name the type of bond formed when a molecule of BF_3 reacts with an F^- ion. Explain how this bond is formed.

Type of bond

.....
.....

Explanation

.....
.....

(2 marks)

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Michael Brown

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