
AQA A LEVEL PHYSICS PAPER 1

1. **What is meant by work function (2):** Minimum energy required by an electron to escape from (1) a metal surface (1)
2. **What is meant by ionisation energy? (2):** Minimum energy required to remove an electron from an atom (1) from the ground state (1)
3. **Why do emitted electrons have a kinetic energy that varies up to a maximum value? (4):** Energy of a photon is fixed/constant (1)
Energy is required for electrons to overcome work function ϕ (1)
Electrons deeper into/below the surface of the metal require more energy to overcome work function and therefore have less KE (1)
Maximum KE = Energy of photon - work function ϕ (1)
4. **Explain how (bright ring) diagram is evidence of the wave-particle duality of electrons?:** Diffraction pattern maxima/minima (bright and dark areas) (1) Diffraction is a property of waves- proves wave-behaviour of electrons (1)
5. **What is meant by the ground state?:** The lowest energy state of an atom
6. **Explain the difference between excitation and ionisation (2):** An electron receives a discrete amount of energy for BOTH (1)
Excitation promotes an electron to a higher energy level
Ionisation is when an electron receives enough energy to leave the atom
7. **Explain why only photons of certain frequencies cause excitation (4):** discrete energy levels (1)
Electrons need to absorb a DISCRETE amount of energy to move to a higher energy level (1)

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1. Understanding the eBook Aqa Physics Empa Questions 2015
 - The Rise of Digital Reading Aqa Physics Empa Questions 2015
 - Advantages of eBooks Over Traditional Books
2. Identifying Aqa Physics Empa Questions 2015
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Aqa Physics Empa Questions 2015
 - User-Friendly Interface
4. Exploring eBook Recommendations from Aqa Physics Empa Questions 2015
 - Personalized Recommendations
 - Aqa Physics Empa Questions 2015 User Reviews and Ratings
 - Aqa Physics Empa Questions 2015 and Bestseller Lists
5. Accessing Aqa Physics Empa Questions 2015 Free and Paid eBooks
 - Aqa Physics Empa Questions 2015 Public Domain eBooks
 - Aqa Physics Empa Questions 2015 eBook Subscription Services
 - Aqa Physics Empa Questions 2015 Budget-Friendly Options
6. Navigating Aqa Physics Empa Questions 2015 eBook Formats

- ePub, PDF, MOBI, and More
- Aqa Physics Empa Questions 2015 Compatibility with Devices
- Aqa Physics Empa Questions 2015 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Aqa Physics Empa Questions 2015
 - Highlighting and Note-Taking Aqa Physics Empa Questions 2015
 - Interactive Elements Aqa Physics Empa Questions 2015
- 8. Staying Engaged with Aqa Physics Empa Questions 2015
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Aqa Physics Empa Questions 2015
- 9. Balancing eBooks and Physical Books Aqa Physics Empa Questions 2015
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Aqa Physics Empa Questions 2015
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Aqa Physics Empa Questions 2015
 - Setting Reading Goals Aqa Physics Empa Questions 2015
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Aqa Physics Empa Questions 2015
 - Fact-Checking eBook Content of Aqa Physics Empa Questions 2015
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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