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Terms in this set (29)

monatomic ions	ions that consist of a single atom with a positive or negative charge resulting from the loss or gain of one or more valence electrons
binary compound	compound that is composed of 2 elements
acid	a compound that contains one or more hydrogen atom(s) and produces hydrogen cations when dissolved in water

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base	ions when dissolved in water
law of definite proportions	states that in samples of any chemical compound, the masses of the elements are always in the same proportions.
law of multiple proportions	whenever the same 2 elements form more than one compound, the different masses of one element that combine with the same mass of the other element are in the ratio of small whole numbers

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