

Persuasive Writing

What is an exposition?

An exposition is a piece of writing used to persuade or convince a reader of a writer's point of view or opinion. When we read these texts, we often see an argument being presented. The argument is either for or against a particular topic or theme. These types of texts are not used for entertainment. They are a persuasive type of text, and they are used to persuade or convince.

What is the structure of an exposition?

Title

This tells the reader exactly what the text is about. The title should be direct and to the point.

Introduction

The introduction should be clear and it should state your opinion. Your aim is to grab the reader's attention, and to make them want to read your argument.

1. Include a statement of position on what you believe about the topic. This statement of position should introduce the argument.
2. Provide a brief overview of the important points you will be presenting in your text. You may also want to give some background information about your topic.

Body

The body is a series of ordered and clear arguments. Each of your arguments will be presented in separate paragraphs.

1. Include a topic sentence at the beginning of each paragraph.
2. Organise your paragraphs using time sequence words like: firstly, secondly, furthermore, finally, to conclude.
3. Keep it clear, concise, and to the point. Do not add in any new information in that you did not highlight in your introduction.

Conclusion

The conclusion simply sums up your text in a powerful way. It is very similar to the structure of your introduction.

1. Restate your position as powerfully as possible.
2. Sum up your arguments and remember to not include any new information.

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A Guide To Persuasive Essays A Guides To Writing 5:

Handbook on Developing Curriculum Materials for Teachers Gerald Bailey, Tara Baillargeon, Cara D. Barragree, Ann Elliott, Raymond Doswell, 2010-02-01 This book provides an essential resource for educators and museum professionals who wish to develop education focused eMuseums that feature motivational standards based curriculum for diverse learners The book is divided into three sections Section 1 Planning Developing and Evaluating eMuseums guides the reader through the stages of planning creating and evaluating a user centered eMuseum This section provides an overview of the process of planning creating and evaluating an eMuseum giving small and medium sized museums the framework and guidance needed to create an eMuseum Section 2 Museum and Public School Partnerships A Step by Step Guide for Creating Standards Based Curriculum Materials in High School Social Studies is the second section This section includes how to a form a partnership b create standards based curriculum materials and c provides curriculum material evaluation strategies Section 3 Developing Accessible Museum Curriculum A Handbook for Museum Professionals and Educators Educators in both museums and schools are faced with the task of delivering content to patrons with increasingly diverse interests skills and learning needs This section outlines specific strategies that can be applied to curriculum to expand its application to broader audiences This section includes a content presentation b content process and c content product Throughout the book materials created from the Negro Leagues Baseball Museum NLBM and Kansas State University KSU partnership are included as product examples

Teaching Writing With Rubrics Laura A. Flynn, Ellen M. Flynn, 2004-03-29 What a gift I used your book from the first day Using your guidelines I am now developing my own units and rubrics Rebecca Kennerly Student Resident Teacher Program University of New Mexico Take the guesswork out of writing instruction and assessment The step by step instructions scheduling guidelines lesson plans rubric assessments and reproducible forms in this guide are the result of years of research planning and fine tuning in the classroom Inspiring student creativity while addressing standards and meeting the needs of every student has never been so easy Special features include Step by step instructions on how to set up a classroom writing program Eleven comprehensive classroom tested lesson plans Methods for adapting rubrics for high school students gifted students English Language Learners and students with learning disabilities Strategies for evaluating every step of the writing process from pre writing to final draft How to incorporate Six Trait Writing Discover the excellent writing your students can produce when provided with a rubric and challenged to do their best Once you have used the strategies and materials in this guide you may wonder how you ever got along without them Expanding the Writing Process with Elaboration Joyce Caskey, 2002-02-14 Equip students with the necessary skills for expanding and elaborating their ideas so they can write well structured paragraphs and well supported essays Ready-To-Use Writing Proficiency Lessons and Activities Carol H. Behrman, 2003-12-29 An experienced teacher and author Carol Behrman helps students develop the sound foundation they need to become proficient writers throughout their lives Included are a variety of easy to

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ \text{aq} + \text{Cr}_2\text{O}_7^{2-} \text{aq} \rightarrow 2\text{Cr}^{3+} \text{aq} + 7\text{H}_2\text{O}$

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web jan 29 2020 the ionic form of this reaction is $2\text{Na}^+ \text{aq} + 2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} + \text{SO}_4^{2-} \text{aq} \rightarrow 2\text{Na}^+ \text{aq} + \text{SO}_4^{2-} \text{aq} + \text{CuCl}_2 \text{s}$ the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web mar 11 2014 solution ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ the spectator ions are Cl^- and Na^+
 net ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g} \rightarrow \text{H}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ here is a video on spectator ions
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web the six NO_3^- ions and the six Na^+ ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2 \text{s}$

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web video answer you know spectator ions are the ions present on the both sides of the equation indicate cancels out in the total ionic equations so you know in a case chemical reactions are described with chemical equations which symbolically present the starting materials end products

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web net ionic equations an equation for a reaction in solution showing only those particles that are directly involved in the chemical change spectator ions an ion that is not directly involved in a chemical reaction an ion that does not change oxidation number or composition during a reaction products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+ \text{aq} + \text{OH}^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l}$ the spectator ions in $\text{KOH} \text{aq}$ and $\text{HNO}_3 \text{aq}$ are $\text{K}^+ \text{aq}$ and $\text{NO}_3^- \text{aq}$ how to identify and list the

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