

Chapter 10: Chemical analysis

Knowledge organiser

Pure and impure

In chemistry, a **pure** substance contains a single element or compound that is not mixed with any other substances.

Pure substances melt and boil at specific temperatures.

An **impure** substance contains more than one type of element or compound in a **mixture**.

Impure substances melt and boil at a range of temperatures.

Formulations

Formulations are examples of mixtures. They have many different components (substances) that make them up in very specific proportions (amounts compared to each other).

Scientists spend a lot of time trying to get the right components in the right proportions to make the most useful product.

Formulations include fuels, cleaning agents, paints, alloys, fertilisers, and foods.



Testing gases

Common gases can be identified using the follow tests:

Gas	What you do	What you observe if gas is present
hydrogen	hold a lighted splint near the gas	hear a squeaky pop
oxygen	hold a glowing splint near the gas	splint re-lights
carbon dioxide	bubble the gas through limewater	the limewater turns milky (cloudy white)
chlorine	hold a piece of damp litmus near the gas	litmus turns the litmus white

Chromatography

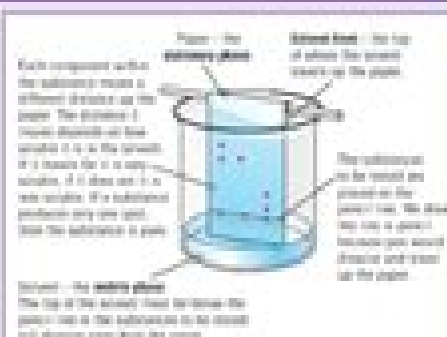
Chromatography is a method to separate different components in a mixture. It is set up as shown here, with a piece of paper in a beaker containing a small amount of solvent.

The **R_f value** is a ratio of how far up the paper a certain spot moves compared to how far the **solvent** has travelled.

$$R_f = \frac{\text{distance moved by substance}}{\text{distance moved by solvent}}$$

It will always be a number between 0 and 1.

The **R_f value** depends on the solvent and the temperature, and different substances will have different **R_f values**. The **R_f values** for particular substances can be used to identify a substance.



Separating mixtures

Mixtures can be separated by:

- Filtration** - separates insoluble solids from a liquid
- Crystallisation** - evaporates a solvent (liquid) leaving the solute (solid)
- Simple distillation** - separates solvent from a solution as long as the solvent has a lower boiling point than the solute
- Fractional distillation** - separates two or more substances from a liquid mixture



Key terms

Make sure you can write a definition for these key terms.

chromatography

formulation
solvent

impure
solvent front

mobile phase
stationary phase

pure

R_f value

A Natural Approach To Chemistry Chapter 10 Review Answers

**James Silk Buckingham, John
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A Natural Approach To Chemistry Chapter 10 Review Answers:

Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity

Fundamentals of Medicinal Chemistry Gareth Thomas, 2004-04-20 Provides a concise introduction to the chemistry of therapeutically active compounds written in a readable and accessible style The title begins by reviewing the structures and nomenclature of the more common classes of naturally occurring compounds found in biological organisms An overview of medicinal chemistry is followed by chapters covering the discovery and design of drugs pharmacokinetics and drug metabolism The book concludes with a chapter on organic synthesis followed by a brief look at drug development from the research stage through to marketing the final product The text assumes little in the way of prior biological knowledge relevant biology is included through biological topics examples and the Appendices Incorporates summary sections examples applications and problems Each chapter contains an additional summary section and solutions to the questions are provided at the end of the text Invaluable for undergraduates studying within the chemical pharmaceutical

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Mery,2012-12-06 In the 1990s the world community has arrived at a particularly in developing countries and in economic historical turning point Global issues the decline in biodiversity in transition These three organizations have of biological diversity climate change the fate of different backgrounds and focuses but have found forest peoples fresh water scarcity desertification it relevant and rewarding to their core operations to deforestation and forest degradation have come collaborate in WFSE activities The intention of to dominate the public and political debate about these organizations is to continue supporting the forestry In the economic sphere forest industries WFSE research and developing the mutual collaboration have assumed global dimensions oration The World Forests Society and Environment In the year 2000 WFSE took on a new challenge Research Program WFSE is a response by the re extending its research network to involve five new search community to this globalization The WFSE Associate Partners the Center for International slogan Globalization calls for global research re Forestry Research CIFOR in Indonesia the Center reflects both the means and the end of the program for Research and Higher Education on Natural The program is involved in promoting and executing Resources of Tropical America CATIE in Costa Rica ing research in different parts of the world and Rica the International Center for Research in Agro through its publications and communications net Forestry ICRAF in Kenya the World Forestry work linking researchers worldwide

Molecular Structure Norman L. Allinger,2010-12-15 A guide to analyzing the structures and properties of organic molecules Until recently the study of organic molecules has traveled down two disparate intellectual paths the experimental or physical method and the computational or theoretical method Working somewhat independently of each other these disciplines have guided research for decades but they are now being combined efficiently into one unified strategy Molecular Structure delivers the essential fundamentals on both the experimental and computational methods then goes further to show how these approaches can join forces to produce more effective analysis of the structure and properties of organic compounds by Looking at experimental structures electron neutron X ray diffraction and microwave spectroscopy as well as computational structures ab initio semi empirical molecular orbital and molecular mechanics calculations Discussing various electronic effects particularly stereoelectronic effects including hyperconjugation negative hyperconjugation the Bohlmann and anomeric effects and how and why these cause changes in structures and properties of molecules Illustrating complex carbohydrate effects such as the gauche effect the delta two effect and the external anomeric torsional effect Covering hydrogen bonding the CH bond and how energies especially heats of formation can be affected Using molecular mechanics to tie all of these things together in the familiar language of the organic chemist valence bond pictures Authored by a founding father of computational chemistry Molecular Structure broadens the scope of the subject by serving as a pioneering guide for workers in the fields of organic biological and computational chemistry as they explore new possibilities to advance their discoveries This work will also be of interest to many of those in tangential or dependent fields including medicinal and pharmaceutical chemistry and pharmacology *The Athenaeum* James Silk Buckingham, John Sterling, Frederick Denison

Maurice, Henry Stebbing, Charles Wentworth Dilke, Thomas Kibble Hervey, William Hepworth Dixon, Norman Maccoll, Vernon Horace Rendall, John Middleton Murry, 1871 Clinical Neuropsychology Study Guide and Board Review Kirk Stucky, Kirk J. Stucky, Michael W. Kirkwood, Jacobus Donders, Christine Liff, 2020 Clinical Neuropsychology Study Guide and Board Review Second Edition provides an easy to study volume with sample questions and recommended readings that are specifically designed to help individuals prepare for the ABCN written examination This book can also be used as a teaching tool for graduate students and trainees at various levels The format is geared toward exam preparation Information is provided in a concise outlined manner with liberal use of bullets boxes illustrations and tables The guide also includes hundreds of mock exam questions and many recommended readings **Aquatic Ecosystems: Interactivity of Dissolved Organic Matter** Stuart Findlay, Robert L. Sinsabaugh, 2003 Overviews of the source supply and variability of DOM surveys of the processes that mediate inputs to microbial food webs and syntheses consolidating research findings provide a comprehensive review of what is known of DOM in freshwater This book will be important to anyone interested in understanding the fundamental factors associated with DOM that control aquatic ecosystems BOOK JACKET

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