Guided Reading Activity

netw rks

The Ancient Greeks

Lesson 3 Classical Greece

Review Questions

develop their own empire.

- Detail: After an extended war against __\PERSIA______, the Greeks, Spartans, and Athenians united to defeat the Persians at __SALAMIS__ and again at ___PLATAEA__.
- B. Main Idea: The Classical period of Athenian and Greek history lasted from and 461 s.c. to 429 s.c. Historians have called this period the Age of Pericles, in which Athens reached the height of its power and brilliance.
 - Detail: Under Athenian leadership, the ___DELIAN __ League liberated virtually all
 of the Greek states in the __AEGEAN _____ from Persian control.
 - Detail: Pericles created a direct __DEMOCRACY______ that included every male citizen and used __OSTRACISM_____ to protect against ambitious politicians.
 - Detail: Most people in the ancient world owned _SLAVES , who worked in the FIELDS , in the home, or on government PUBLIC CONSTRUCTION _ projects.
 - Detail: Athenian women were ____CITIZENS___, but they were excluded from __PUBLIC (political)__ life and expected to work chiefly at ___home___.
- C. Main Idea: After they defeated Persia, the Greek city-states returned to rivalry and war among themselves.
 - Detail: After they defeated the Persians, Athens and the Delian League formed the
 __ATHENIAN____ Empire, while Sparta and its supporters formed a rival
 PELOPONNESIAN__ League.
 - 2. Detail: The Great _PELOPONNESIAN WAR_ between Sparta and ____ATHENS__ weakened the major Greek states and allowed them to fall prey to the growing _MACEDONIAN____ empire.

Answers American History Guided Activity 3 3

Joyce Oldham Appleby, Alan Brinkley, James M. McPherson

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web for the first five problems you need to use the equation that says that the molarity of a solution is equal to the number of moles of solute divided by the number of liters of solution in this problem simply solve using the molarity equation to find that the concentration of the solution is 10 m

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web molality differs from molarity only in the denominator while molarity is based on the liters of solution molality is based on the kilograms of solvent concentrations expressed in molality are used when studying properties of solutions related to vapor pressure and temperature changes

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web answer the mass percentage of an element of a solution is defined as the mass of the solute in grammes present in 100 g of the solution it s written like this begin array l mass of a component frac mass of component in solution total mass of solution x 100 end array

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