



Gizmos

Name: Date:

Student Exploration: Periodic Trends

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: atomic radius, electron affinity, electron cloud, energy level, group, ion, ionization energy, metal, nonmetal, nucleus, period, periodic trends, picometer, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. On the image at right, the two magnets are the same. Which paper clip would be harder to remove?

2. Which magnet would be most likely to attract additional paper clips?

3. What is the relationship between the thickness of the book and the ability of the magnet to hold on to and attract paper clips?



Gizmo Warm-up

Just as the thickness of a book changes how strongly a magnet attracts a paper clip, the size of an atom determines how strongly the **nucleus** attracts electrons. In the *Periodic Trends Gizmo*, you will explore this relationship and how it affects the properties of different elements.

The **atomic radius** is a measure of the size of the **electron cloud**, or the region where electrons can be found. To begin, check that **H** (hydrogen) is selected in **Group 1** on the left. Turn on **Show ruler**. To measure the radius, drag one end of the ruler to the proton in the nucleus and the other end to the electron. Click **Save radius** to record the value.



1. What is the radius of hydrogen?

Notice that the radius is measured in **picometers** (pm). A picometer is one trillionth of a meter.

2. On the right side of the Gizmo, select **Li**. Connect the right side of the ruler to the outermost electron, or **valence electron**. What is the radius of lithium?

Answers For Gizmo Prisms And Culinders Activity A

RD Boyd



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Table of Contents Answers For Gizmo Prisms And Culinders Activity A

1. Understanding the eBook Answers For Gizmo Prisms And Culinders Activity A
 - The Rise of Digital Reading Answers For Gizmo Prisms And Culinders Activity A
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Gizmo Prisms And Culinders Activity A
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Gizmo Prisms And Culinders Activity A
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Gizmo Prisms And Culinders Activity A
 - Personalized Recommendations
 - Answers For Gizmo Prisms And Culinders Activity A User Reviews and Ratings
 - Answers For Gizmo Prisms And Culinders Activity A and Bestseller Lists
5. Accessing Answers For Gizmo Prisms And Culinders Activity A Free and Paid eBooks
 - Answers For Gizmo Prisms And Culinders Activity A Public Domain eBooks
 - Answers For Gizmo Prisms And Culinders Activity A eBook Subscription Services
 - Answers For Gizmo Prisms And Culinders Activity A Budget-Friendly Options

6. Navigating Answers For Gizmo Prisms And Culinders Activity A eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Gizmo Prisms And Culinders Activity A Compatibility with Devices
 - Answers For Gizmo Prisms And Culinders Activity A Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Gizmo Prisms And Culinders Activity A
 - Highlighting and Note-Taking Answers For Gizmo Prisms And Culinders Activity A
 - Interactive Elements Answers For Gizmo Prisms And Culinders Activity A
8. Staying Engaged with Answers For Gizmo Prisms And Culinders Activity A
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Gizmo Prisms And Culinders Activity A
9. Balancing eBooks and Physical Books Answers For Gizmo Prisms And Culinders Activity A
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Gizmo Prisms And Culinders Activity A
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers For Gizmo Prisms And Culinders Activity A
 - Setting Reading Goals Answers For Gizmo Prisms And Culinders Activity A
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers For Gizmo Prisms And Culinders Activity A
 - Fact-Checking eBook Content of Answers For Gizmo Prisms And Culinders Activity A
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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