

Name: _____

Date: _____

Student Exploration: pH Analysis

Vocabulary: acid, acidic, alkaline, base, indicator, neutral, pH

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. **Acids** are substances that produce hydrogen ions (H^+) when dissolved in water. Lemon juice is an example of an acid.

- A. What does lemon juice taste like? sour, tangy taste
- B. What does it feel like if lemon juice gets in your eye? Sting and burn

2. **Bases** are substances that produce hydroxide ions (OH^-) when dissolved in water. Hand soap is an example of a base.

- A. What does soap feel like? Slippery, smooth
- B. What does soap taste like? Bitter
- C. What does it feel like if soap gets in your eye? Burns

Gizmo Warm-up

The strength of an acid or base is measured on the **pH** scale. The term "pH" is short for "potential of hydrogen." It is a measure of how many excess H^+ ions there are in a solution. The pH scale runs from 0 to 14, with 0 representing the highest concentration of hydrogen ions. **Acidic** substances have a pH below 7, while **alkaline** substances (bases) have a pH above 7. Pure water has a pH of 7 and is considered **neutral**.

The *pH Analysis* Gizmo allows you to find the pH of a variety of liquids. In the Gizmo, check that the **Substance in the tube** is **Ammonia**, and click **Test**. Wait until the animation is finished.



1. **Indicators** change color in acids or bases. What is the color of the pH paper? Dark blue
2. Compare the paper to the **pH color chart**. What is the pH of ammonia? 11
3. Is ammonia acidic or alkaline? Alkaline

Answers Ph Gizmo

Ed Sobey, Ph.D.



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Table of Contents Answers Ph Gizmo

1. Understanding the eBook Answers Ph Gizmo
 - The Rise of Digital Reading Answers Ph Gizmo
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers Ph Gizmo
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers Ph Gizmo
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers Ph Gizmo
 - Personalized Recommendations
 - Answers Ph Gizmo User Reviews and Ratings
 - Answers Ph Gizmo and Bestseller Lists
5. Accessing Answers Ph Gizmo Free and Paid eBooks
 - Answers Ph Gizmo Public Domain eBooks
 - Answers Ph Gizmo eBook Subscription Services
 - Answers Ph Gizmo Budget-Friendly Options

6. Navigating Answers Ph Gizmo eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers Ph Gizmo Compatibility with Devices
 - Answers Ph Gizmo Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers Ph Gizmo
 - Highlighting and Note-Taking Answers Ph Gizmo
 - Interactive Elements Answers Ph Gizmo
8. Staying Engaged with Answers Ph Gizmo
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers Ph Gizmo
9. Balancing eBooks and Physical Books Answers Ph Gizmo
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers Ph Gizmo
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers Ph Gizmo
 - Setting Reading Goals Answers Ph Gizmo
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers Ph Gizmo
 - Fact-Checking eBook Content of Answers Ph Gizmo
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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