Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



- What type of charge does a proton have? Positive (+1)
- What type of charge does a neutron have? Neutral (0)
- What type of charge does an electron have? Negative (-1)
- Which two subatomic particles are located in the nucleus of an atom?
 Protons and neutrons
- 8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
 35
- 9. What is the atomic number of the atom in the diagram above?
- What is the atomic mass/mass number of the atom in the diagram above?
 (11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number
- 11. How many protons are in the nucleus of an atom with an atomic number of 15?
 15 *number of protons is the same as the atomic number
- How many electrons are in the nucleus of an atom with an atomic number of 20?
- How many neutrons are in the nucleus of an atom with an atomic number of 25? (use Periodic Table for mass)
 Neutrons, atomic mass: 54.938 ~ 55 atomic number: 25....55-25=30
- 14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
 Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.
- How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
 Neutrons= 11 (atomic mass: 36 - 25 :atomic number)

Answers For Atomic Structure

Oswaal Editorial Board

Answers For Atomic Structure:

Answers For Atomic Structure Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Answers For Atomic Structure**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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