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web read standard reflux on page 201 of the organic chem lab survival manual prelab question 1 trimyristin is a triglyceride what are triglycerides prelab question 2 how do microwaves heat microwave procedure add approximately 2-5 g of ground nutmeg record exact mass and 15 ml of diethyl ether to microwave vessel

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web in this week's lab experiment the purpose of this lab is to new techniques isolate the chemical trimyristin from the common spice nutmeg this lab requires five techniques simple distillation vacuum filtration melting point determination reflux and extraction

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web hatice kocataş bursa uludağ university faculty of arts and sciences department of chemistry 16059 bursa turkey abstract in the present work a surfactant assisted convenient extraction method was developed for the isolation of trimyristin from nutmeg

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web natural products chemistry the isolation of trimyristin from nutmeg over 40 of the medicinal chemicals used throughout the developed world today were originally isolated from natural sources these sources include flowering plants fungi bacteria and to a lesser extent animals especially marine animals

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web nov 1 2021 the skill of isolating trimyristin will be useful for students to obtain natural whitening raw materials derived from nutmeg various studies on the isolation can be used in natural products

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web oct 18 2019 extraction is made a lot easier when extracting trimyristin from nutmeg because of its large quantity in nutmeg the nutmeg is moved from a solid to a liquid phase which is then used for extracting and isolating the trimyristin
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web trimyristin compound from nutmeg seed trimyristin compounds contained in nutmeg seed is very important to be isolated because it has various benefits trimyristin along with myristic acid myristicin and elemicin have antioxidant anticonvulsant analgesic anti inflammatory antidiabetic antibacterial and anti fungal activity

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web aug 4 2008 the purpose of this experiment was extract isolate and purify the natural product trimyristin from the spice nutmeg the trimyristin was extracted by using the solvent diethyl ether and

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web trimyristin is found in the fixed oil of nutmeg the fixed oil comprises approximately 24 40 of the nutmeg seed trimyristin comprises 73 of the fixed oil overall trimyristin should have percent recovery of 18 29 1 figure 1 shows how trimyristin is triester formed from the dehydration reaction between glycerol and myristic acid oh

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web purpose the purpose of this lab is to extract trimyristin from nutmeg as well as synthesizing myristic acid from the extracted trimyristin using base hydrolysis and naoh acidification using hcl is necessary to characterize the product by finding the melting point

lab 4 extraction of trimyristin from nutmeg university of toronto - Sep 07 2022

web in this lab trimyristin is extracted from ground nutmeg occurring in many vegetable oils as well as myristic acid trimyristin consists of long saturated hydrocarbon chains and it is relatively nonpolar thus using a relatively nonpolar solvent trimyristin can be easily extracted by the concept of like dissolves like

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web the desired components of solids are dissolved in a solvent then the remaining solid is filtered so that the selected element and solvent are left over in this lab the trimyristin was dissolved into the solvent and the remainder of the nutmeg was filtered out

lab experiment 3 trimyristin extraction from nutmeg docsity - Feb 12 2023

web may 1 2011 experiment 3 1 february 2010 grade a small scale extraction of trimyristin from nutmeg introduction the purpose of this experiment is to collect trimyristin from an amount of nutmeg provided trimyristin is a type of triglyceride containing the fatty acid myristic acid also known as tetradecanoic acid

nutmeg lab final report isolation and purification of trimyristin - May 03 2022

web final report isolation and purification of trimyristin from nutmeg 2 purpose the purpose of this experiment is to isolate a chemical in nutmeg called trimyristin through extraction and to perform ir spectroscopy on this sample to determine its functional groups chemical structures signatures