

Electrostatics

Electro — charge

Static — rest

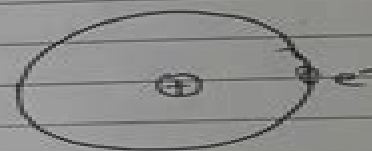
Charges —

① positive charges

② negative charges

→ In atom $\frac{A}{Z}X$ there is

no. of proton = no. of electron = Z



* electron having -ve charge & its magnitude $-1.6 \times 10^{-19} \text{C}$
by gain of electron by atom

positive charge if loss of electron from atom
its magnitude $= 1.6 \times 10^{-19} \text{C}$.

* gain of electron

no. of electron $>$ no. of proton (In atom)



* loss of electron

no. of electron $<$ no. of proton



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Xiang Xie



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Table of Contents 2014 June Exemplars Grade 12 Caps Physics

1. Understanding the eBook 2014 June Exemplars Grade 12 Caps Physics
 - The Rise of Digital Reading 2014 June Exemplars Grade 12 Caps Physics
 - Advantages of eBooks Over Traditional Books
2. Identifying 2014 June Exemplars Grade 12 Caps Physics
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook 2014 June Exemplars Grade 12 Caps Physics
 - User-Friendly Interface
4. Exploring eBook Recommendations from 2014 June Exemplars Grade 12 Caps Physics
 - Personalized Recommendations
 - 2014 June Exemplars Grade 12 Caps Physics User Reviews and Ratings

- 2014 June Exemplars Grade 12 Caps Physics and Bestseller Lists
- 5. Accessing 2014 June Exemplars Grade 12 Caps Physics Free and Paid eBooks
 - 2014 June Exemplars Grade 12 Caps Physics Public Domain eBooks
 - 2014 June Exemplars Grade 12 Caps Physics eBook Subscription Services
 - 2014 June Exemplars Grade 12 Caps Physics Budget-Friendly Options
- 6. Navigating 2014 June Exemplars Grade 12 Caps Physics eBook Formats
 - ePub, PDF, MOBI, and More
 - 2014 June Exemplars Grade 12 Caps Physics Compatibility with Devices
 - 2014 June Exemplars Grade 12 Caps Physics Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 2014 June Exemplars Grade 12 Caps Physics
 - Highlighting and Note-Taking 2014 June Exemplars Grade 12 Caps Physics
 - Interactive Elements 2014 June Exemplars Grade 12 Caps Physics
- 8. Staying Engaged with 2014 June Exemplars Grade 12 Caps Physics
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 2014 June Exemplars Grade 12 Caps Physics
- 9. Balancing eBooks and Physical Books 2014 June Exemplars Grade 12 Caps Physics
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 2014 June Exemplars Grade 12 Caps Physics
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 2014 June Exemplars Grade 12 Caps Physics
 - Setting Reading Goals 2014 June Exemplars Grade 12 Caps Physics
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 2014 June Exemplars Grade 12 Caps Physics
 - Fact-Checking eBook Content of 2014 June Exemplars Grade 12 Caps Physics
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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