Matter—Properties and Changes

Lesson 2.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical	mass		physical			
density	prope	rties	substance			
Matter is anythin	ng with (1)	MASS	and volume.	A		
(2) _SUBSTANCE	I	is a form of n	natter with a uniform and une	changing		
composition. Substi	ances have speci	ific, unchanging (3)	PROPERTIES	that can be		
observed. Substance	es have both phy	vsical and chemical;	properties. (4)			
PHYSICAL		propertie	s can be observed without ch	anging a substance's		
chemical compositi	on. Color, hardr	ness, and				
(5)DENSITY_		are examples. Other properties cannot be observed without				
changing the compo	sition of a subs	tance. These are call	ed (6)CHEMICAL			
properties. An exam	iple is the tender	ncy of iron to form r	ust when exposed to air.			
Label each proper	ty as either <i>pl</i> (y	sical or chemical.				
CHEM	7. Cb	7. Chemical formula H ₂ O				
CHEM	8. Fo	8. Forms green carbonate when exposed to moist air				
CHEM	9. Re	9. Remains unchanged when in the presence of nitrogen				
PHYS	10.	Colorless				
PHYS	11.	Solid at norma	I temperatures and pressures			
CHEM	12.	Ability to com	bine with another substance			
PHYS	13.	Melting point				
PHYS	14.	Liquid at norm	al temperatures and pressure	s		
PHYS	15.	Boiling point it	s 100°C			

Conducts electricity

PHCY'S.

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