
CHEMISTRY
SCIENCE Paper – 2

(Two hours)

Answers to this Paper must be written on the paper provided separately.

*You will **not** be allowed to write during the first 15 minutes.*

This time is to be spent in reading the Question Paper.

The time given at the head of this paper is the time allowed for writing the answers.

Section I is compulsory. Attempt **any four** questions from **Section II**.

The intended marks for questions or parts of questions are given in brackets [].

SECTION I (40 Marks)

*Attempt **all** questions from this Section*

Question 1

- (a) Choose the most appropriate answer.
- (i) Which of the following is a common characteristic of a covalent compound?
- A High melting point.
 - B Conducts electricity when it is in the molten state.
 - C Consists of molecules.
 - D Always soluble in water.
- (ii) Ammonium hydroxide will produce a reddish brown precipitate when added to a solution of :
- A CuSO_4
 - B $\text{Zn}(\text{NO}_3)_2$
 - C FeSO_4
 - D FeCl_3
- (iii) A salt which in solution gives a bluish white precipitate with NaOH solution and a white precipitate with BaCl_2 solution is:
- A CuSO_4
 - B FeSO_4
 - C $\text{Fe}_2(\text{SO}_4)_3$
 - D CuCl_2

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