

QUESTION 5

- 5.1
- 5.1.1 Aufbau diagram ✓✓ (2)
 - 5.1.2 Sulphur ✓✓ (2)
 - 5.1.3 $1s^2 2s^2 2p^4 3s^2 3p^2$ ✓✓ (2)
 - 5.1.4 Group 16 ✓ (2)
 - Period 3 ✓
 - 5.1.5 The element has 6 valence electrons ✓ and belong to the third horizontal row of the periodic table. ✓ (2)
 - 5.1.5 Non-metal ✓ (1)
- 5.2.
- 5.2.1 5 and 1, respectively (1)
 - 5.2.2 $\begin{array}{c} \text{Y} \text{ : } \text{X} \text{ : } \text{Y} \\ \vdots \\ \text{Y} \end{array}$ ✓ (2)
 - 5.2.4 N ✓ and H ✓, or P and H (2)
- [18]

QUESTION 6

- 6.1
- (a) 24 ✓
 - (b) 12 ✓
 - (c) Na^+ ✓
 - (d) 10 ✓
 - (e) 12 ✓
 - (f) 17 ✓
 - (g) 19 ✓
- 6.2 Relative atomic mass = $(0,76 \times 35)$ ✓ + $(0,24 \times 37)$ ✓
= 35,48 ✓ (3)
- 6.3 Isotopes ✓✓ (2)
- [12]

QUESTION 7

- 7.1 Sodium hydroxide (NaOH) ✓
Hydrogen ✓ or $\text{H}_2(\text{g})$ (2)
- 7.2 $2\text{Mg}(\text{s}) + \text{O}_2(\text{g}) \rightarrow 2\text{MgO}(\text{s})$ ✓✓ (bal) (3)
- 7.3 Sodium + water → sodium hydroxide + water (5)
- $2\text{Na}(\text{s}) + 2\text{H}_2\text{O}(\text{l}) \rightarrow 2\text{NaOH}(\text{aq}) + \text{H}_2(\text{g})$ ✓ ✓ (Bal)
- 7.4 Because they react vigorously with air to form oxides ✓ and with water to form hydroxides. ✓ (2)
- 7.5 It is an inert colourless gas ✓
It is unreactive ✓ (2)
- 7.6 Noble gases (1)
- 7.7 Because they don't react, it was not easy to discover them, so no one knew they existed. ✓✓ (2)

2014 Grade11 Physical Sciences Exemplar

Toby Karten



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Science Blogging Christie Wilcox, Bethany Brookshire, Jason G. Goldman, 2016-03-01 Here is the essential how to guide for communicating scientific research and discoveries online ideal for journalists researchers and public information officers looking to reach a wide lay audience Drawing on the cumulative experience of twenty seven of the greatest minds in scientific communication this invaluable handbook targets the specific questions and concerns of the scientific community offering help in a wide range of digital areas including blogging creating podcasts tweeting and more With step by step guidance and one stop expertise this is the book every scientist science writer and practitioner needs to approach the Wild West of the Web with knowledge and confidence

Common Core Standards and English Language Arts Grades 6 -12: Strategies for Student Success Toby Karten, 2013-01-01 The tri fold laminated reference guide Common Core Standards English Language Arts Strategies for Student Success Grades 6 12 by Toby Karten presents an at a glance overview of the CCSS for English Language Arts ELA for students in grades 6 12 It is intended to help middle and high school teachers understand the organization and application of the standards for diverse students including those with special needs Topics covered in the guide include Challenges of the cross disciplinary design of CCSS for ELA at the secondary level College and career readiness CCR CCR anchor standards for reading literature RL and informational text RI CCR anchor standards for writing W CCR anchor standards for speaking and listening SL CCR anchor standards for language L Differentiating instruction for students at different reading levels Models for measuring and evaluating text complexity Strategies for helping students with disabilities achieve ELA standards The CCSS for ELA raise the bar to ensure students master the reading writing speaking and listening and language skills they need to be college and career ready in literacy They will yield effective outcomes for all students including those with disabilities if educators creatively and consistently embrace them and connect them to all of their students

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Table of Contents 2014 Grade11 Physical Sciences Exemplar

1. Understanding the eBook 2014 Grade11 Physical Sciences Exemplar
 - The Rise of Digital Reading 2014 Grade11 Physical Sciences Exemplar
 - Advantages of eBooks Over Traditional Books
2. Identifying 2014 Grade11 Physical Sciences Exemplar
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 2014 Grade11 Physical Sciences Exemplar
 - User-Friendly Interface
4. Exploring eBook Recommendations from 2014 Grade11 Physical Sciences Exemplar
 - Personalized Recommendations
 - 2014 Grade11 Physical Sciences Exemplar User Reviews and Ratings
 - 2014 Grade11 Physical Sciences Exemplar and Bestseller Lists

5. Accessing 2014 Grade11 Physical Sciences Exemplar Free and Paid eBooks
 - 2014 Grade11 Physical Sciences Exemplar Public Domain eBooks
 - 2014 Grade11 Physical Sciences Exemplar eBook Subscription Services
 - 2014 Grade11 Physical Sciences Exemplar Budget-Friendly Options
6. Navigating 2014 Grade11 Physical Sciences Exemplar eBook Formats
 - ePub, PDF, MOBI, and More
 - 2014 Grade11 Physical Sciences Exemplar Compatibility with Devices
 - 2014 Grade11 Physical Sciences Exemplar Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 2014 Grade11 Physical Sciences Exemplar
 - Highlighting and Note-Taking 2014 Grade11 Physical Sciences Exemplar
 - Interactive Elements 2014 Grade11 Physical Sciences Exemplar
8. Staying Engaged with 2014 Grade11 Physical Sciences Exemplar
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 2014 Grade11 Physical Sciences Exemplar
9. Balancing eBooks and Physical Books 2014 Grade11 Physical Sciences Exemplar
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 2014 Grade11 Physical Sciences Exemplar
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 2014 Grade11 Physical Sciences Exemplar
 - Setting Reading Goals 2014 Grade11 Physical Sciences Exemplar
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 2014 Grade11 Physical Sciences Exemplar
 - Fact-Checking eBook Content of 2014 Grade11 Physical Sciences Exemplar
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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