

Matter—Properties and Changes

Section 3.1 Properties of Matter

pages 70–75

Problem-Solving Lab

1. Explain why the flow of a compressed gas must be controlled for practical and safe use.

The flow of compressed gas must be controlled to control the amount and the rate at which gas is released.

2. Predict what would happen if the valve on a full tank of compressed gas were suddenly opened all the way or if the tank were accidentally punctured.

Without the regulator device, the gas would rush out of the tank with a force powerful enough to transform the tank into a dangerous, uncontrolled projectile.

Section 3.1 Assessment

page 75

1. Create a table that describes the three common states of matter in terms of their shape, volume, and compressibility.

	Volume	Shape	Compressibility
Solid	Definite	Definite	Incompressible
Liquid	Definite	Takes shape of container and fills container to the extent of its own volume	Virtually incompressible
Gas	Fills volume of container	Takes shape of container	Compressible

2. Describe the characteristics that identify a sample of matter as being a substance.

The sample of matter must have a uniform and unchanging composition to be a substance.

3. Classify each of the following as a physical or chemical property.

- a. Iron and oxygen form rust.
chemical

- b. Iron is more dense than aluminum.

physical

- c. Magnesium burns brightly when ignited.

chemical

- d. Oil and water do not mix.

physical

- e. Mercury melts at -39°C .

physical

4. **Organize** Create a chart that compares physical and chemical properties. Give two examples for each type of property.

The chart should make clear that physical properties can be observed without changing the composition of the sample, which is not the case for chemical properties. Mass and density are examples of physical properties. Fermentation and rusting are examples of chemical properties.

Section 3.2 Changes in Matter

pages 76–79

Practice Problems

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5. Use the data in the table to answer the following questions.

Aluminum and Liquid Bromine Reaction

	Before Reaction	After Reaction
Aluminum	10.3 g	0.0 g
Liquid bromine	100.0 g	8.5 g
Compound	0.0 g	

How many grams of bromine reacted? How many grams of compound were formed?

amount of bromine that reacted =
 $100.0\text{ g} - 8.5\text{ g} = 91.5\text{ g}$

amount of compound formed =
 $100.0\text{ g} + 10.3\text{ g} - 8.5\text{ g} = 101.8\text{ g}$

Answers Chemistry Matter And Change

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Answers Chemistry Matter And Change:

Chemistry: Core Concepts, 3rd Edition Allan Blackman, Daniel Southam, Gwendolyn Lawrie, Natalie Williamson, Christopher Thompson, 2024-01-09 The third edition of Chemistry Core Concepts Blackman et al has been developed by a group of leading chemistry educators for students entering university with little or no background in chemistry Available as a full colour printed textbook with an interactive eBook code this title enables every student to master concepts and succeed in assessment Lecturers are supported with an extensive and easy to use teaching and learning package

Physical and Chemical Changes (eBook) Edward P. Ortleb, Richard Cadice, 1993-09-01 This book presents a program of basic studies in physical and chemical changes of matter The definition of matter is presented along with explanations of states and properties of matter Topics include atoms molecules elements compounds mixtures solutions symbols and formulas Each of the twelve teaching units in this book is introduced by a color transparency print books or PowerPoint slide eBooks that emphasizes the basic concept of the unit and presents questions for discussion Reproducible student pages provide reinforcement and follow up activities The teaching guide offers descriptions of the basic concepts to be presented background information suggestions for enrichment activities and a complete answer key

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decreases by 2.b. ... answer the following questions.Nuclear ... 25.2 Nuclear Transformations | Lecture notes Chemistry These nuclei decay by turning a neutron into a pro- ton to emit a beta particle (an electron) from the nucleus. This process is known as beta emission. It ... 60 s - 1 min SECTION 25.2 NUCLEAR TRANSFORMATIONS. 1. Write a nuclear equation for the following radioactive processes. a. alpha decay of francium-208 $^{208}\text{Fr} \rightarrow \text{b} \dots$ The Sorrows of Travel: a Novel: John Breon ASIN, B0000CJEJQ. Publisher, Peter Davies; First Edition (January 1, 1956). Language, English. Hardcover, 222 pages. Item Weight, 1.74 pounds. The sorrows of travel, by Edward Abbey 20th century American authors Arizona Biography Edward Abbey Fire lookouts Man-woman relationships Relations with women United States ... The sorrows of travel,: A novel: Breon, John ASIN, B0007E5L1W. Publisher, Putnam; First Edition (January 1, 1955). Hardcover, 250 pages. Item Weight, 1.1 pounds. The Sorrows Of Travel by John Breon - AbeBooks Hardcover - Peter Davies - 1956 - Condition: Good - Dust Jacket Included - 1956. First Published. 221 pages. Pictorial dust jacket over beige cloth. Sorrows of Travel by Breon, John - 1955 The book is about Paris, young Americans after the ww2, enjoying literary Paris and the life. Dust jacket shows wear around edges, has tears top and bottom of ... The Sorrows of Travel a novel uncorrected proof The Sorrows of Travel a novel [uncorrected proof]. Breon, John. London: Peter Davies, 1956. Paperback. 222p., very good uncorrected proof copy ... The Sorrows of Travel | Cincinnati & Hamilton County ... The Sorrows of TravelThe Sorrows of Travel. Breon, JohnBreon, John. Title rated 0 out of 5 stars, based on 0 ratings (0 ratings). The Sorrows of Others May 9, 2023 — In Tolstoy Together: 85 Days of War and Peace, Yiyun Li invites you to travel with her through Tolstoy's novel—and with fellow readers ... The Best of Edward Abbey The Sorrows of Travel. When I think of travel I think of certain women I have known. So many of my own journeys have been made in pursuit of love. In pursuit ...