



Name: _____ Date: _____

Student Exploration: Nuclear Decay

Background: Alpha particles, beta particles, gamma rays, and neutrons are particles that are emitted from unstable nuclei. Alpha particles consist of two protons and two neutrons, beta particles consist of one electron or one positron, gamma rays consist of high-energy photons, and neutrons consist of one neutron.

Prior Knowledge Questions (What do you already know about this topic?)

1. What are some properties of atoms, isotopes, and radioactive materials? (What do you know about them?)

Particle	Composition	Charge	Approximate mass
Alpha	Two protons and two neutrons	+2	4 u
Beta	One electron or one positron	-1 or +1	1/1836 u
Gamma	High-energy photon	0	0

1. The **atomic number** of an atom is equal to the number of protons in the nucleus.

2. Alpha particles have 2 protons and 2 neutrons. What is the mass number of the parent? (What is the mass number of the daughter?)

3. The **mass number** of an atom is equal to the sum of the number of protons and the number of neutrons in the nucleus. If the parent has a mass number of 238, what is the mass number of the daughter?

Alpha Decay

Alpha decay occurs when an unstable nucleus emits an alpha particle, which consists of two protons and two neutrons. This process is called **alpha decay**.

Background: The decay of a radioactive nucleus is a random process. It is not possible to predict when a particular nucleus will decay.

However, we can predict the half-life of a radioactive material. The half-life is the time it takes for half of the material to decay. The half-life of a radioactive material is a constant.



1. Alpha particles have a mass number of 4 and an atomic number of 2. What is the mass number of the parent? (What is the mass number of the daughter?)

2. Alpha particles have a charge of +2. What is the charge of the parent? (What is the charge of the daughter?)

Background: Alpha particles are emitted from the nucleus of a radioactive material. They have a mass number of 4 and a charge of +2.



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IM Harris



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