

SCHAU Unit 4 - Quiz – Equilibrium

Name: _____ Period: _____

120 _____

120 _____

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120 _____

$$K_{eq} = \frac{[products]}{[reactants]} \quad K_{eq} = \frac{[C]^c [D]^d}{[A]^a [B]^b}$$

$$K_{eq} = \frac{[C]^c [D]^d}{[A]^a [B]^b} \quad K_{eq} = \frac{[C]^c [D]^d}{[A]^a [B]^b}$$

120 _____ 120 _____ 120 _____ 120 _____ 120 _____

Question 1

1. Hydrogen sulfide, H_2S , is a slightly poisonous gas. At 100°C, an equilibrium mixture contains 0.020 moles of hydrogen, H_2 , and 0.010 moles of hydrogen sulfide, with an equilibrium pressure of 0.10 atm. The reaction is:



If the volume of the 0.100 atm at 100°C is kept constant, what is the partial pressure of H_2 at equilibrium?

- 0.020 atm
 Since H_2 is a diatomic gas.
 Since H_2S is a diatomic gas.
 H_2 is a gas.
- 0.010 atm, at equilibrium.
- 0.020 atm.
 And the reaction has reached equilibrium at 100°C.
- 0.010 atm.

	H_2S	H_2	S_2
Initial (moles)			
Change (moles)			
Final (moles)			

120 _____

Answers For Sch4u Unit 4

JG Myers



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