## Chem - Covalent Bonding: Questions & Answers

molecules 
Ans - seperate groups of atoms with a formula, bonded together

intermolecular bonds ✓ Ans - bonds between molecules (dispersion, dipole-dipole, hydrogen bonding) determines physical properties

intramolecular bonds 
Ans - bonds between atoms in a molecule (covalent, ionic eg.)

covalent bonds 
Ans - non metals share electrons to complete the octet rule, an atom forms as many bonds as it needs eg. carbon needs 4 electrons so forms 4 bonds

why do non metals share electrons not transfer? 

Ans - non metals have a relatively high electronegativity so competing to attract each others electron, shares instead

electron dot diagrams - Ans - lewis diagram shows valence electrons (bonding pairs) and lone pairs

valence structure 
✓ Ans - each pair of bonding electrons is shown as a line, still represents dots as lone pairs. 1 line=2 electrons

bonding pairs and lone pairs 
Ans - bonding pairs; shared electrons in bond 
lone pairs; non bonding pairs

double covalent bond ✓ Ans - rigid, two pairs of electrons (4 electrons) are shared, represented as a double line in valence structures eg. o2

# **Answers For Chem Think Covalent Bonding**

**Leallyn Burr Clapp** 

## **Answers For Chem Think Covalent Bonding:**

Chemistry of the Covalent Bond Leallyn Burr Clapp,1957

The book delves into Answers For Chem Think Covalent Bonding. Answers For Chem Think Covalent Bonding is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Answers For Chem Think Covalent Bonding, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
  - Chapter 1: Introduction to Answers For Chem Think Covalent Bonding
  - Chapter 2: Essential Elements of Answers For Chem Think Covalent Bonding
  - Chapter 3: Answers For Chem Think Covalent Bonding in Everyday Life
  - Chapter 4: Answers For Chem Think Covalent Bonding in Specific Contexts
  - ∘ Chapter 5: Conclusion
- 2. In chapter 1, the author will provide an overview of Answers For Chem Think Covalent Bonding. This chapter will explore what Answers For Chem Think Covalent Bonding is, why Answers For Chem Think Covalent Bonding is vital, and how to effectively learn about Answers For Chem Think Covalent Bonding.
- 3. In chapter 2, this book will delve into the foundational concepts of Answers For Chem Think Covalent Bonding. The second chapter will elucidate the essential principles that need to be understood to grasp Answers For Chem Think Covalent Bonding in its entirety.
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- 6. In chapter 5, the author will draw a conclusion about Answers For Chem Think Covalent Bonding. The final chapter will summarize the key points that have been discussed throughout the book.

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