

### **Ionic and Covalent Naming Practice **Answers****

*For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.*

- |     |                           |                                  |
|-----|---------------------------|----------------------------------|
| 1)  | $\text{Na}_2\text{CO}_3$  | <b>sodium carbonate</b>          |
| 2)  | $\text{P}_4\text{O}_{10}$ | <b>tetraphosphorus decaoxide</b> |
| 3)  | $\text{N}_2\text{O}_5$    | <b>dinitrogen pentoxide</b>      |
| 4)  | $\text{FeSO}_4$           | <b>iron (II) sulfate</b>         |
| 5)  | $\text{SiO}_2$            | <b>silicon dioxide</b>           |
| 6)  | $\text{FeCl}_3$           | <b>iron (III) chloride</b>       |
| 7)  | $\text{CoBr}_2$           | <b>cobalt (II) bromide</b>       |
| 8)  | $\text{B}_2\text{H}_4$    | <b>diboron tetrahydride</b>      |
| 9)  | $\text{CO}$               | <b>carbon monoxide</b>           |
| 10) | $\text{PCl}_3$            | <b>phosphorus trichloride</b>    |

*For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.*

- |     |                        |                                                      |
|-----|------------------------|------------------------------------------------------|
| 11) | dinitrogen trioxide    | <b><math>\text{N}_2\text{O}_3</math></b>             |
| 12) | nitrogen dioxide       | <b><math>\text{NO}_2</math></b>                      |
| 13) | sulfur pentoxide       | <b><math>\text{SO}_5</math></b>                      |
| 14) | lithium acetate        | <b><math>\text{LiC}_2\text{H}_3\text{O}_2</math></b> |
| 15) | phosphorus trifluoride | <b><math>\text{PF}_3</math></b>                      |
| 16) | vanadium (V) oxide     | <b><math>\text{V}_2\text{O}_5</math></b>             |
| 17) | aluminum hydroxide     | <b><math>\text{Al}(\text{OH})_3</math></b>           |
| 18) | zinc sulfide           | <b><math>\text{ZnS}</math></b>                       |
| 19) | silicon tetrafluoride  | <b><math>\text{SiF}_4</math></b>                     |
| 20) | silver phosphate       | <b><math>\text{Ag}_3\text{PO}_4</math></b>           |

# Answers Ionic And Covalent Bond Practice Chemfiesta

**Primrose Kitten**

A decorative red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the 'Primrose Kitten' text.

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**Chemical Bonding** M.S. Sethi & M. Satake, 2010 Contents Chemical Bonding I Basic Concepts Chemical Bonding II Additional Aspects Intermolecular Force and Crystal Structures

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then applied to a variety of covalent molecules with discussions of bond lengths and angles and hence molecular shapes Extensive comparisons of valence bond theory and VSEPR theory with molecular orbital theory are included Metallic bonding is related to electrical conduction and semi conduction The energetics of ionic bond formation and the transition from ionic to covalent bonding is also covered

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**The Chemical Bond** Gernot Frenking,Sason Shaik,2014-06-13 This is the perfect complement to Chemical Bonding Across the Periodic Table by the same editors who are two of the top scientists working on this topic each with extensive experience and important connections within the community The resulting book is a unique overview of the different approaches used for describing a chemical bond including molecular orbital based valence bond based ELF AIM and density functional based methods It takes into account the many developments that have taken place in the field over the past few decades due to the rapid advances in quantum chemical models and faster computers

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