

Chemical Formula Writing Worksheet Solutions

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl₂", as shown.

	zinc	iron (II)	iron (III)	gallium	silver	lead (IV)
chloride	ZnCl₂	FeCl₂	FeCl₃	GaCl₃	AgCl	PbCl₄
acetate	Zn(C₂H₃O₂)₂	Fe(C₂H₃O₂)₂	Fe(C₂H₃O₂)₃	Ga(C₂H₃O₂)₃	Ag C₂H₃O₂	Pb(C₂H₃O₂)₄
nitrate	Zn(NO₃)₂	Fe(NO₃)₂	Fe(NO₃)₃	Ga(NO₃)₃	AgNO₃	Pb(NO₃)₄
oxide	ZnO	FeO	Fe₂O₃	Ga₂O₃	Ag₂O	PbO₂
nitride	Zn₃N₂	Fe₃N₂	FeN	GaN	Ag₃N	Pb₃N₄
sulfate	ZnSO₄	FeSO₄	Fe₂(SO₄)₃	Ga₂(SO₄)₃	Ag₂SO₄	Pb(SO₄)₂

Write the formulas for the following compounds:

- 1) copper (II) chloride **CuCl₂**
- 2) lithium acetate **LiC₂H₃O₂**
- 3) vanadium (III) selenide **V₂Se₃**
- 4) manganese (IV) nitride **Mn₃N₄**
- 5) beryllium oxide **BeO**
- 6) sodium sulfate **Na₂SO₄**
- 7) aluminum arsenide **AlAs**
- 8) potassium permanganate **KMnO₄**
- 9) chromium (VI) cyanide **Cr(CN)₆**
- 10) tin (II) sulfite **SnSO₃**
- 11) vanadium (V) fluoride **VF₅**
- 12) ammonium nitrate **NH₄NO₃**

Answers For Ionic Formula Writing

Scott C. Dulebohn

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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