

Matter—Properties and Changes

Lesson 2.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical
density

mass
properties

physical
substance

Matter is anything with (1) _____ **MASS** _____ and volume. A (2) _____ **SUBSTANCE** _____ is a form of matter with a uniform and unchanging composition. Substances have specific, unchanging (3) _____ **PROPERTIES** _____ that can be observed. Substances have both physical and chemical properties. (4) _____ **PHYSICAL** _____ properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) _____ **DENSITY** _____ are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) _____ **CHEMICAL** _____ properties. An example is the tendency of iron to form rust when exposed to air.

Label each property as either *physical* or *chemical*.

- | | |
|-------------------------|---|
| _____ CHEM _____ | 7. Chemical formula H_2O |
| _____ CHEM _____ | 8. Forms green carbonate when exposed to moist air |
| _____ CHEM _____ | 9. Remains unchanged when in the presence of nitrogen |
| _____ PHYS _____ | 10. Colorless |
| _____ PHYS _____ | 11. Solid at normal temperatures and pressures |
| _____ CHEM _____ | 12. Ability to combine with another substance |
| _____ PHYS _____ | 13. Melting point |
| _____ PHYS _____ | 14. Liquid at normal temperatures and pressures |
| _____ PHYS _____ | 15. Boiling point is $100^{\circ}C$ |
| _____ PHYS _____ | 16. Conducts electricity |

3 Properties Of Matter Study Guide

Scott C. Dulebohn



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