

# Acids and Bases

How do acids and bases behave in water?

## Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

## Model 1 – Arrhenius Acids and Bases

| Common Name for Aqueous Solution | Chemical Formula                                      | Found in...       | Tastes... | Turns Litmus Paper... | Conducts Electricity? | Acid or Base |
|----------------------------------|-------------------------------------------------------|-------------------|-----------|-----------------------|-----------------------|--------------|
| Acetic acid                      | $\text{HC}_2\text{H}_3\text{O}_2(\text{aq})$          | Vinegar           | Sour      | Red                   | Yes                   | Acid         |
| Benzoic acid                     | $\text{HC}_7\text{H}_5\text{O}_2(\text{aq})$          | Food preservative | Sour      | Red                   | Yes                   | Acid         |
| Phosphoric acid                  | $\text{H}_3\text{PO}_4(\text{aq})$                    | Soda pop          | Sour      | Red                   | Yes                   | Acid         |
| Hydrochloric acid                | $\text{HCl}(\text{aq})$                               | Stomach acid      | Sour      | Red                   | Yes                   | Acid         |
| Citric acid                      | $\text{H}_2\text{C}_6\text{H}_7\text{O}_6(\text{aq})$ | Citrus fruits     | Sour      | Red                   | Yes                   | Acid         |
| Ascorbic acid                    | $\text{H}_2\text{C}_6\text{H}_7\text{O}_6(\text{aq})$ | Vitamin C         | Sour      | Red                   | Yes                   | Acid         |
| Magnesium hydroxide              | $\text{Mg}(\text{OH})_2(\text{aq})$                   | Milk of magnesia  | Bitter    | Blue                  | Yes                   | Base         |
| Aluminum hydroxide               | $\text{Al}(\text{OH})_3(\text{aq})$                   | Antacids          | Bitter    | Blue                  | Yes                   | Base         |
| Barium hydroxide                 | $\text{Ba}(\text{OH})_2(\text{aq})$                   | Lubricants        | POISON    | Blue                  | Yes                   | Base         |
| Sodium hydroxide                 | $\text{NaOH}(\text{aq})$                              | Drain cleaner     | POISON    | Blue                  | Yes                   | Base         |

- Refer to Model 1.
  - What is the common chemical name for Vitamin C?
  - Is Vitamin C classified as an acid or a base?
- Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.
- Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
- Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.

# 32 Acids And Bases Key

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