

COURSE STRUCTURE
CLASS-XI (THEORY) (2024-25)

Time: 3 Hours

Total Marks 70

S.No	UNIT	No. of Periods	Marks
1	Some Basic Concepts of Chemistry	12	7
2	Structure of Atom	14	9
3	Classification of Elements and Periodicity in Properties	8	6
4	Chemical Bonding and Molecular Structure	14	7
5	Chemical Thermodynamics	16	9
6	Equilibrium	14	7
7	Redox Reactions	6	4
8	Organic Chemistry: Some basic Principles and Techniques	14	11
9	Hydrocarbons	12	10
	TOTAL		70

Nd Puc New Syllabus Cbse Chemistry Manual

**William MACPHERSON (Professor of
Chemistry, Ohio State University, and
HENDERSON (William Edwards))**

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Chemistry Lab Manual Class XII | follows the latest CBSE syllabus and other State Board following the CBSE Curriculum.

Mr. Rohit Manglik, Mr. Kaushalesh Dwivedi, 2022-08-04 With the NEP and expansion of research and knowledge has changed the face of education to a great extent In the Modern times education is not just constricted to the lecture method but also includes a practical knowledge of certain subjects This way of education helps a student to grasp the basic concepts and principles Thus trying to break the stereotype that subjects like Physics Chemistry and Biology means studying lengthy formulas complex structures and handling complicated instruments we are trying to make education easy fun and enjoyable

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Practical/Laboratory Manual Chemistry Class - XI Er. Meera Goyal, 2021-05-29

- 1 Basic Laboratory Techniques
 - 1 To cut a glass tube or glass rod
 - 2 To bend the glass rod at an angle
 - 3 To draw a glass jet from a glass tube
 - 4 To bore a cork and fit a glass tube into it
- 2 Characterisation and Purification of Chemical Substances
 - 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique Viva Voce
 - 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method Viva Voce
 - 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample
 - 4 To prepare the pure crystals of copper sulphate from the given crude sample
 - 5 To prepare pure crystals of benzoic acid from a given impure sample Viva Voce
- 3 Measurement of pH Values
 - 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
 - 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration
 - 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
 - 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH
 - 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ Viva Voce
- 4 Chemical Equilibrium
 - 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
 - 2 To study the shift in equilibrium between Co^{2+} and Cl^- ions by changing the concentrations of either of the ions Viva Voce
- 5 Quantitative Analysis
 - 1 To prepare M/10 oxalic acid solution by direct weighing method
 - 2 To prepare M/10 solution of sodium carbonate by direct weighing method
 - 3 To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid
 - 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a

standard N 10 or M 20 sodium carbonate solution Viva Voce 6 Qualitative Analysis Analysis of Anions Analysis of Cations Viva Voce 7 Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne s test 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne s test Viva Voce INVESTIGATORY PROJECTS 1 Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions Viva Voce 2 Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 o purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids IViva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce Arihant CBSE Chemistry Term 2 Class 12 for 2022 Exam (Cover Theory and MCQs) Arshdeep Kaur,Aditya Jangid,2021-11-20 With newly introduced 2 Term Examination Pattern CBSE has eased out the pressure of preparation of subjects and cope up with lengthy syllabus Introducing Arihant s CBSE TERM II 2022 Series the first of its kind that gives complete emphasize on the rationalize syllabus of Class 9th to 12th The all new CBSE Term II 2022 Chemistry of Class 12th provides explanation and guidance to the syllabus required to study efficiently and succeed in the exams The book provides topical coverage of all the chapters in a complete and comprehensive manner Covering the 50% of syllabus as per Latest Term wise pattern 2021 22 this book consists of 1 Complete Theory in each Chapter covering all topics 2 Case Based Short and Long Answer Type Question in each chapter 3 Coverage of NCERT NCERT Exemplar Board Exams Questions 4 Complete and Detailed explanations for each question 5 3 Practice papers base on entire Term II Syllabus Table of Content Electrochemistry Chemical Kinetics Surface Chemistry The d and f Block Coordination Compounds Aldehydes Ketones and Carboxylic Acids Amines Practice Papers 1 3 **Comprehensive Practical Chemistry XII** Dr. N . K. Verma,B. K. Vermani,Dr. Neera Verma,K . K. Rehani,2011-11 **A Manual of Degree Practical Chemistry** Mehdi Al Kausor,Sharmistha Chakraborty,Ruhul Amin Bepari,2024-04 It gives us immense pleasure to place this The book entitled A Manual of Degree Practical Chemistry before the student community written in accordance with the recent syllabus based on NEP 2020 The aim of writing this book is to guide the students having Chemistry Major and Minor Course for performing the experiments to be carried out in Chemistry laboratory Keeping in view the requirement of the

students and the teachers this book has been written to cover the topics with the desired limits of the prescribed syllabus. The Authors have made an attempt to make this subject and concepts of experiment simpler and more understandable to the budding pharmacy student readers and hope that the book will be useful and meets the requirements of students and academicians. Any suggestions and corrections if any towards further improvement will be thankfully acknowledged and highly appreciated.

Arihant CBSE Chemistry Term 2 Class 11 for 2022 Exam (Cover Theory and MCQs) Aditya Jangid, Arshdeep Kaur, 2021-11-20. With the newly introduced 2 Term Examination Pattern CBSE has eased out the pressure of preparation of subjects and cope up with lengthy syllabus. Introducing Arihant's CBSE TERM II 2022 Series the first of its kind that gives complete emphasis on the rationalized syllabus of Class 9th to 12th. The all new CBSE Term II 2022 Chemistry of Class 11th provides explanation and guidance to the syllabus required to study efficiently and succeed in the exams. The book provides topical coverage of all the chapters in a complete and comprehensive manner. Covering the 50% of syllabus as per Latest Term wise pattern 2021-22, this book consists of 1 Complete Theory in each Chapter covering all topics, 2 Case Based Short and Long Answer Type Question in each chapter, 3 Coverage of NCERT NCERT Exemplar Board Exams Questions, 4 Complete and Detailed explanations for each question, 5 3 Practice papers based on the entire Term II Syllabus. Table of Content: States of Matter, Gases and Liquids, Chemical Thermodynamics, Equilibrium, s Block Element, Hydrocarbons, Practice Papers 1-3.

Practical/Laboratory Manual Chemistry Class XI based on NCERT guidelines by Dr. S. C. Rastogi & Er. Meera Goyal Dr. S. C. Rastogi, Er. Meera Goyal, 2020-06-23. An Excellent Book in Accordance with the latest syllabus for Class 11 Prescribed by CBSE NCERT and Adopted by Various State Education Boards. A Basic Laboratory Techniques: 1 To cut a glass tube or glass rod, 2 To bend the glass rod at an angle, 3 To draw a glass jet from a glass tube, 4 To bore a cork and fit a glass tube into it. B Characterisation and Purification of Chemical Substances: 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique, 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method, 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample, 4 To prepare the pure crystals of copper sulphate from the given crude sample, 5 To prepare pure crystals of benzoic acid from a given impure sample. C Measurement of pH Values: 1 To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper, 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration, 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper, 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH , 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+ . D Chemical Equilibrium: 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions, 2 To study the shift in equilibrium between Co^{2+} , H_2O , $6Cl^-$ and Cl_2 ions by changing the concentrations of either of the ions. E Quantitative Analysis: 1 To prepare M/10 oxalic acid

solution by direct weighing method 2 To prepare M 10 solution of sodium carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution F Qualitative Analysis 1 Analysis of Anions 2 Analysis of Cations G Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne s test 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne s test INVESTIGATORY PROJECTS A Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions B Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method C Water Analysis 1 To test the hardness of different water samples D Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap E Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper F Analysis of Fruits and Vegetable Juices 1 To analyse the fruit and vegetable juices for the constituent present in them G Rate of Evaporation 1 To study the rate of evaporation of different liquids H Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Log Antilog Table Chemistry & Cdr & Study Guide & Sol Manual & Stud Companion & Media Activities B University Loretta Jones, University Lecturer in Physical Chemistry Fellow and Tutor P W Atkins, 2000-05-30 *A Course in General Chemistry [with] Laboratory Manual* William MACPHERSON (Professor of Chemistry, Ohio State University, and HENDERSON (William Edwards)), 1913

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