

CHEMISTRY PAPER-II

OBJECTIVE

TIME ALLOWED: 20 Minutes

MAXIMUM MARKS: 17

Note: You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill that bubble in front of that question number, on bubble sheet. Use marker or pen to fill the bubbles. Cutting or filling two or more bubbles will result in zero mark in that question. No credit will be awarded in case BUBBLES are not filled. Do not solve question on this sheet of OBJECTIVE PAPER.

Q.No.1

- (1) Which one of the following does not belong to alkaline earth metals?
(A) *Be* (B) *Ra* (C) *Ba* (D) *Rn*
- (2) Which element forms an ion with charge +3?
(A) *Be* (B) *Al* (C) *C* (D) *Si*
- (3) Oxidation of *NO* in air produces:
(A) *N₂O* (B) *N₂O₃* (C) *N₂O₄* (D) *N₂O₅*
- (4) Chlorine heptaoxide (*Cl₂O₇*) reacts with water to form:
(A) Hypochlorous acid (B) Chloric acid (C) Perchloric acid (D) Chlorine and oxygen
- (5) The anhydride of *HClO₄* is:
(A) *ClO₃* (B) *ClO₂* (C) *Cl₂O₃* (D) *Cl₂O₇*
- (6) Which of the following is a non-typical transition element?
(A) *Cr* (B) *Mn* (C) *Zn* (D) *Fe*
- (7) Ethers show the phenomenon of:
(A) Position isomerism (B) Functional group isomerism (C) Metamerism (D) Cis-trans isomerism
- (8) Preparation of vegetable ghee involves:
(A) Halogenation (B) Hydrogenation (C) Hydroxylation (D) Dehydrogenation
- (9) Which of the following acid can be used as a catalyst in Friedal-Crafts reactions?
(A) *AlCl₃* (B) *HNO₃* (C) *BeCl₂* (D) *NaCl*
- (10) For which mechanisms, the first step involved is the same?
(A) *E₁* and *E₂* (B) *E₂* and *S_N2* (C) *S_N1* and *E₂* (D) *E₁* and *S_N1*
- (11) Which compound is more soluble in water?
(A) *C₂H₅OH* (B) *C₆H₅OH* (C) *CH₃COCH₃* (D) *n*-Hexanol
- (12) Which of the following will have the highest boiling point?
(A) Methanal (B) Ethanal (C) Propanal (D) 2-Hexanone
- (13) Which of the following reagents will react with both aldehydes and ketones?
(A) Grignard's reagent (B) Tollen's reagent (C) Fehling's reagent (D) Benedict's reagent
- (14) Which reagent is used to reduce a carboxylic group to an alcohol?
(A) *H₂/Ni* (B) *H₂/Pt* (C) *NaBH₄* (D) *LiAlH₄*
- (15) Alkanenitriles can be prepared by treating alkyl halide with:
(A) Alcoholic *KOH* (B) Alcoholic *KCN* (C) Aqueous *KOH* (D) Aqueous *KNO₃*
- (16) Most concentrated solid nitrogen fertilizer is:
(A) *NH₃* (B) *(NH₄)₂HPO₄* (C) *(NH₂)₂CO* (D) *NH₄NO₃*
- (17) Mark the correct statement.
(A) *Na⁺* is smaller than *Na* atom (B) *Na⁺* is larger than *Na* atom
(C) *Cl⁻* is smaller than *Cl* atom (D) *Cl⁻* ion and *Cl* atom are equal in size

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Arshad Iqbal



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Douglas Buchanan, James Melrose, A. R. Ross, 2015

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Arshad Iqbal, 2020-04-10 Previously published as A Level Chemistry MCQs Multiple Choice Questions and Answers Quiz Tests with Answer Keys by Arshad Iqbal A Level Chemistry Multiple Choice Questions and Answers MCQs A Level Chemistry quizzes practice tests with answer key provides mock tests for competitive exams to solve 1745 MCQs A Level Chemistry MCQs helps with theoretical conceptual and analytical study for self assessment career tests This book can help to learn and practice A Level Chemistry quizzes as a quick study guide for placement test preparation A level Chemistry Multiple Choice Questions and Answers MCQs is a revision guide with a collection of trivia quiz questions and answers on topics Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry ionic equilibria lattice energy moles and equations nitrogen and sulfur organic and nitrogen compounds periodicity polymerization rates of reaction reaction kinetics redox reactions and electrolysis states of matter transition elements to enhance teaching and learning A level Chemistry Quiz Questions and Answers also covers the syllabus of many competitive papers for admission exams of different universities from chemistry textbooks on chapters Alcohols and Esters Multiple Choice Questions 27 MCQs Atomic Structure and Theory Multiple Choice Questions 37 MCQs Benzene Chemical Compound Multiple Choice Questions 41 MCQs Carbonyl Compounds Multiple Choice Questions 29 MCQs Carboxylic Acids and Acyl Compounds Multiple Choice Questions 27 MCQs Chemical Bonding Multiple Choice Questions 213 MCQs Chemistry of Life Multiple Choice Questions 29 MCQs Electrode Potential Multiple Choice Questions 62 MCQs Electrons in Atoms Multiple Choice Questions 53 MCQs Enthalpy Change Multiple Choice Questions 45 MCQs Equilibrium Multiple Choice Questions 50 MCQs Group IV Multiple Choice Questions 53 MCQs Groups II and VII Multiple Choice Questions 180 MCQs Halogenoalkanes Multiple Choice Questions 33 MCQs Hydrocarbons Multiple Choice Questions 53 MCQs Introduction to Organic Chemistry Multiple Choice Questions 52 MCQs Ionic Equilibria Multiple Choice Questions 56 MCQs Lattice Energy Multiple Choice Questions 33 MCQs Moles and Equations Multiple Choice Questions 50 MCQs Nitrogen and Sulfur Multiple Choice Questions 89 MCQs Organic and Nitrogen Compounds Multiple Choice Questions 54 MCQs Periodicity Multiple Choice Questions 202 MCQs Polymerization Multiple Choice Questions 36 MCQs Rates of Reaction Multiple Choice Questions 39 MCQs Reaction Kinetics Multiple Choice Questions 52 MCQs Redox Reactions and Electrolysis Multiple Choice Questions 55 MCQs States of Matter Multiple Choice Questions 66 MCQs Transition Elements Multiple Choice Questions 29 MCQs The chapter Alcohols and Esters MCQs covers topics of introduction to alcohols and alcohols reactions The chapter Atomic Structure and Theory MCQs covers topics of atom facts elements and atoms number of nucleons protons electrons and neutrons The chapter Benzene Chemical Compound MCQs

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