

34	a) How $^{35}\text{Cl}_{17}$ and $^{37}\text{Cl}_{17}$ are related to each other? b) If both of these forms of Cl occurs in nature as 75% and 25% respectively. Calculate the average atomic mass of Chlorine atom. c) Draw the labelled schematic atomic structure of Chlorine atom having $Z=17$. Or a) Distinguish between the properties of electrons, protons, and neutrons with respect to position, charge and mass. b) Write the valency and atomic symbol of the following element: i. Sodium ($Z=11$) ii. Aluminium ($Z=13$) iii. Potassium ($Z=19$) iv. Calcium ($Z=20$)	5
35	a) Plastids are present only in the plant. Which types of plastids are present in : i. Petals; ii. Leaves; iii. fruit pulp b) What happen if chloroplast is removed from the plant. Explain. Or To test the effect of the concentration of the solution on the cell a student placed some plant cell and animal cell (RBC) for 15 minute in the hypertonic solution and then shifted them into hypotonic solution. Answer the following question: a) What is hypotonic solution. b) What will the expected observation in the above situation. c) what is the role of cell wall in the cell in above situation.	5
36	a) Explain universal law of gravitation also write its mathematical expression. b) Write three importance of this law. Or Give reason: a) carry a bag with broader strap is easy than a thin strap bag. b) why sharp knife cut easy in comparison with the blunt knife. c) camel can run in a desert easily.	5
SECTION-E Q.no.37 to 39 are case based / source- based questions with 2 to 3 short sub - parts.		

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Table of Contents 2014 Grade1tourism Question Paper

1. Understanding the eBook 2014 Grade1tourism Question Paper
 - The Rise of Digital Reading 2014 Grade1tourism Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying 2014 Grade1tourism Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 2014 Grade1tourism Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from 2014 Grade1tourism Question Paper
 - Personalized Recommendations
 - 2014 Grade1tourism Question Paper User Reviews and Ratings

- 2014 Grade1tourism Question Paper and Bestseller Lists
- 5. Accessing 2014 Grade1tourism Question Paper Free and Paid eBooks
 - 2014 Grade1tourism Question Paper Public Domain eBooks
 - 2014 Grade1tourism Question Paper eBook Subscription Services
 - 2014 Grade1tourism Question Paper Budget-Friendly Options
- 6. Navigating 2014 Grade1tourism Question Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - 2014 Grade1tourism Question Paper Compatibility with Devices
 - 2014 Grade1tourism Question Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 2014 Grade1tourism Question Paper
 - Highlighting and Note-Taking 2014 Grade1tourism Question Paper
 - Interactive Elements 2014 Grade1tourism Question Paper
- 8. Staying Engaged with 2014 Grade1tourism Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 2014 Grade1tourism Question Paper
- 9. Balancing eBooks and Physical Books 2014 Grade1tourism Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 2014 Grade1tourism Question Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 2014 Grade1tourism Question Paper
 - Setting Reading Goals 2014 Grade1tourism Question Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 2014 Grade1tourism Question Paper
 - Fact-Checking eBook Content of 2014 Grade1tourism Question Paper
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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