

MEMO

Question 1

1.1. List six (6) properties of an ideal primary standard. (6)

- High purity ✓
- Atmospheric stability ✓
- Absence of hydrate water ✓
- Readily available at a modest cost ✓
- Reasonable solubility in the titration medium ✓
- Reasonably large molar mass ✓

1.2. Write down the indicator, which is used in each of the following argentometric titrations and the corresponding end point color.

a) Mohr method (2)

Na_2CrO_4 ✓

Brick red Ag_2CrO_4 precipitate is formed ✓

b) Volhard method (2)

Fe (III) ✓

Red FeSCN^{2+} is formed. ✓

c) Fajans method (2)

Fluorescein ✓

The fluoresceinate ion forms an intensely red silver salt. ✓

Question 2

2.1. Calculate the molar solubility of $\text{Pb}(\text{IO}_3)_2$ in a solution that contains 0.050 M Pb^{2+} . ($K_{sp} = 3.18 \times 10^{-13}$) (7)



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Liying Dong



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