

- Atomic Number
Symbol
Name
Atomic Mass

Atomic number equals
the number of

Protons or Electrons

Atomic mass equals

the number of protons + neutrons



Atomic # = 8
Atomic Mass = 16
# of Protons = 8
# of Neutrons = 8
# of Electrons = 8



Atomic # = 30
Atomic Mass = 105
# of Protons = 30
# of Neutrons = 35
# of Electrons = 30

Atomic # = 5
Atomic Mass = 7
# of Protons = 3
# of Neutrons = 4
# of Electrons = 3

# Silicon 28.086

Atomic # = 14
Atomic Mass = 28
# of Protons = 14
# of Neutrons = 14
# of Electrons = 14

Atomic # = 5
Atomic Mass = 11
# of Protons = 5
# of Neutrons = 5
# of Electrons = 5



Atomic # = 35

Atomic Mass = 80

# of Protons = 35

# of Neutrons = 45

# of Electrons = 35

# **Answer Key To The Atoms Family Atomic Math Challenge**

**Clemens Wendtner** 

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