



Equations

Name _____

Date _____

Write a balanced chemical equation to represent each of the following chemical reactions.

1. iron + sulfur $\rightarrow$ iron(II) sulfide
2. zinc + copper(II) sulfate $\rightarrow$ zinc sulfate + copper
3. silver nitrate + sodium bromide $\rightarrow$ sodium nitrate + silver bromide
4. potassium chlorate (heated) $\rightarrow$ potassium chloride + oxygen
5. water (electricity) $\rightarrow$ hydrogen + oxygen
6. mercury(II) oxide (heated) $\rightarrow$ mercury + oxygen
7. potassium iodide + lead(II) nitrate $\rightarrow$ lead(II) iodide + potassium nitrate
8. aluminum + oxygen $\rightarrow$ aluminum oxide
9. magnesium chloride + ammonium nitrate $\rightarrow$ magnesium nitrate + ammonium chloride
10. iron(III) chloride + ammonium hydroxide $\rightarrow$ iron(III) hydroxide + ammonium chloride
11. sodium peroxide + water $\rightarrow$ sodium hydroxide + oxygen
12. iron(III) oxide + carbon $\rightarrow$ iron + carbon monoxide
13. iron + water $\rightarrow$ hydrogen + iron(III) oxide
14. iron(III) chloride + potassium hydroxide $\rightarrow$ potassium chloride + iron(III) hydroxide
15. aluminum + sulfuric acid $\rightarrow$ aluminum sulfate + hydrogen
16. sodium carbonate + calcium hydroxide $\rightarrow$ sodium hydroxide + calcium carbonate



Answer Key Weston Walch Chemistry

Yan Bai



Answer Key Weston Walch Chemistry:

Chemistry Thomas Kardos, 2003 This book provides a cornerstone for understanding atomic structure chemical bonding chemical reactions the periodic table and more It contains teacher demos and lab activities that stimulate scientific inquiry checked for safety and designed for easy inexpensive use *Top Shelf* Brian Pressley, 2003 Covers chemical formulas and equations chemical reactions structure of atoms the gas laws and more Presents hands on activities as catalysts to fuel student imagination **Practical Chemistry Labs** Leonard Saland, 1989 Features self contained step by step activities using common materials and covering topics from food chemistry to papermaking and electrochemistry Illustrates the connection between the real world and chemistry concepts such as solutions chemistry acids and bases and more Includes teacher notes quizzes and answers to help monitor student progress *Chemistry Problems* David E. Newton, 2001 This edition includes acid base chemistry and thermochemistry Chemistry Problems is the authoritative resource for practice problems covering all the essentials Includes Atomic structure Stoichiometry Solutions chemistry and Electrochemistry Literally thousands of problems in this compendium build proficiency analytical skills and math skills The text includes a complete answer key and reference to applicable web sites 40 Low-Waste, Low-Risk Chemistry Labs David Dougan, 1997 Builds essential process and thinking skills Investigates central chemistry concepts Features procedures for purchase storage use and disposal of chemicals **Resources in Education**, 1997 Curriculum Review, 1983

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