

Experiment 10: Electrochemistry with a Standard Hydrogen Electrode from: *Basic Electrochemistry*, Experiment 10, pp. 45-50

The purpose of this experiment is to determine standard potentials for several other electrodes. In the first part of the experiment, the standard hydrogen electrode (SHE) is used as a reference electrode. The standard potential of the SHE is defined as 0.00 V. The standard potential of the SHE is determined by measuring the potential of the SHE against a standard hydrogen electrode (SHE) at a standard concentration of 1.0 M H^+ and 1.0 atm H_2 gas. The standard potential of the SHE is determined by measuring the potential of the SHE against a standard hydrogen electrode (SHE) at a standard concentration of 1.0 M H^+ and 1.0 atm H_2 gas. The standard potential of the SHE is determined by measuring the potential of the SHE against a standard hydrogen electrode (SHE) at a standard concentration of 1.0 M H^+ and 1.0 atm H_2 gas.

In this experiment, you will determine the standard potentials of several other electrodes. The standard potential of the SHE is defined as 0.00 V. The standard potential of the SHE is determined by measuring the potential of the SHE against a standard hydrogen electrode (SHE) at a standard concentration of 1.0 M H^+ and 1.0 atm H_2 gas.

Objectives:

1. To determine the standard potential of the standard hydrogen electrode (SHE).
2. To determine the standard potential of the standard hydrogen electrode (SHE) against a standard hydrogen electrode (SHE) at a standard concentration of 1.0 M H^+ and 1.0 atm H_2 gas.
3. To determine the standard potential of the standard hydrogen electrode (SHE) against a standard hydrogen electrode (SHE) at a standard concentration of 1.0 M H^+ and 1.0 atm H_2 gas.
4. To determine the standard potential of the standard hydrogen electrode (SHE) against a standard hydrogen electrode (SHE) at a standard concentration of 1.0 M H^+ and 1.0 atm H_2 gas.

Materials:

- Standard hydrogen electrode (SHE)
- Standard hydrogen electrode (SHE)
- Standard hydrogen electrode (SHE)
- Standard hydrogen electrode (SHE)
- Standard hydrogen electrode (SHE)
- Standard hydrogen electrode (SHE)
- Standard hydrogen electrode (SHE)
- Standard hydrogen electrode (SHE)
- Standard hydrogen electrode (SHE)

Procedure:

1. Prepare the standard hydrogen electrode (SHE) at a standard concentration of 1.0 M H^+ and 1.0 atm H_2 gas.
2. Prepare the standard hydrogen electrode (SHE) at a standard concentration of 1.0 M H^+ and 1.0 atm H_2 gas.

Results:

1. The standard potential of the standard hydrogen electrode (SHE) is 0.00 V.
2. The standard potential of the standard hydrogen electrode (SHE) is 0.00 V.

Discussion:

1. The standard potential of the standard hydrogen electrode (SHE) is 0.00 V.
2. The standard potential of the standard hydrogen electrode (SHE) is 0.00 V.

Answers For Lab 6b Heath Chemistry

M Mosston



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