## Student Exploration: Covalent Bonds

**Vocabulary:** covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

There are eight markers in a full set, but Flora and Frank each only have seven markers.
 Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers?

Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

#### Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the Covalent Bonds Gizmo, atoms form bonds in this way.

To begin, check that Fluorine is selected from the Select a substance menu. Click Play ( \*\*) to see the electrons orbiting the nucleus of each atom.



- The outermost electrons in each atom are called valence electrons. How many valence electrons does each fluorine atom have?
- Click Pause (III). Drag an electron from the left atom to the right atom. Click Play. What happens?
- Click Pause, drag an electron from the right atom to the left, and then click Play.
   What happens now?

# **Answers Ofexploration Guide Covalent Bonds**

**Team Prabhat** 

**Answers Of exploration Guide Covalent Bonds:** 

#### Answers Ofexploration Guide Covalent Bonds Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Answers Ofexploration Guide Covalent Bonds**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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